

ANSWER KEY: Ch 17 Acid-base and Solubility Equilibria

Question 1: **17.10** The following table contains the K_a values of several acids. Which acid has the highest K_b value for its conjugate base?

1. The acids, listed in order of increasing K_a values, are: acetic acid ($K_a = 1.8 \times 10^{-5}$), hydrofluoric acid ($K_a = 6.6 \times 10^{-4}$), hydrocyanic acid ($K_a = 4.0 \times 10^{-10}$), and hydrobromic acid ($K_a = 1.0 \times 10^3$).

The acids, listed in order of increasing K_a values, are listed in concordance with acid-base reactions. Which one of the following statements is not true?

- A. The equivalence point and end point will always be exactly the same for any titration.
- B. In the equivalence point, the number of equivalents of the solution being titrated is the same as the number of equivalents of the titrant added.
- C. The end point of a titration is often determined by an indicator color change.
- D. The equivalence point of a titration can be found using a pH meter.

Question 2: **17.11**

1. A 100.0 mL sample of 0.100 M HCl is titrated with 0.100 M NaOH .

How many milliliter of a 0.20 M solution of HCl are needed to neutralize 10 grams of NaOH dissolved in water? (Molar mass of NaOH is 40.00 g/mol)

- A. 100 mL
- B. 1500 mL
- C. 250 mL
- D. 2500 mL

Question 3: **17.12**

1. The concentration of OH^- in a 0.10 M solution of HCl is:

- A. 1.0×10^{-2} mol/L
- B. 1.0×10^{-13} mol/L

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Meera Goyal**

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In a digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**17 Answer Key Solubility Equilibria**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve to the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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