

# Acids and Bases

## Neutralization Reactions Lab



### Reading, Lab, and Practice Bundle (Grades 6-9)

**Acids and Bases**

**What is a Neutralization**

A neutralization reaction is when an acid reacts with a base to form a salt and water.

This type of chemical reaction is important in everyday life - I trusting an upset stomach or to balancing soil pH in gardens.

**What are Acids and Bases**

Acids are substances that taste sour and can have a low pH.

Examples: lemon juice.



**Matching Vocabulary**

Directions: Match each word on the left with the correct definition. Write the correct letter next to each number.

- 1) A substance with a pH greater than 7 that produces  $\text{OH}^-$  ions.
- 2) The product of an acid and base reaction, often found in the form of NaCl.
- 3) A scale from 0 to 14 used to measure how acidic or basic a substance is.
- 4) A chemical that changes color to show if a substance is acidic or basic.
- 5) A substance with a pH less than 7 that produces  $\text{H}^+$  ions.
- 6) A reaction between an acid and a base to form water and salt.

**Matching Vocabulary**

NAME: \_\_\_\_\_

DATE: \_\_\_\_\_

Directions: Match each word on the left with the correct definition or example on the right. Write the correct letter next to each number.

- 1) A substance with a pH greater than 7 that produces  $\text{OH}^-$  ions.
- 2) The product of an acid and base reaction, often found in the form of NaCl.
- 3) A scale from 0 to 14 used to measure how acidic or basic a substance is.
- 4) A chemical that changes color to show if a substance is acidic or basic.
- 5) A substance with a pH less than 7 that produces  $\text{H}^+$  ions.
- 6) A reaction between an acid and a base to form water and salt.

Acid

Base

pH scale

Neutralization

Salt

Indicator

28 Pages Printable & Easy to Use Includes Answer Key

# 19 4 Neutralization Reactions Answer Key

**Charles H. Corwin**



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