

5.4

THE PERIODIC TABLE: ORGANIZING THE ELEMENTS

SECTION REVIEW

Objectives

- Describe the origin of the periodic table
- Identify the position of groups, periods, and the transition metals in the periodic table

Key Terms

- | | | |
|---------------------------|---------------------------|---------------|
| • periodic table | • metals | • nonmetals |
| • periods | • alkali metals | • halogens |
| • periodic law | • alkaline earth metals | • noble gases |
| • group | • transition metals | • metalloids |
| • representative elements | • inner transition metals | |

Part A Completion

Use this completion exercise to check your understanding of the concepts and terms that are introduced in this section. Each blank can be completed with a name, short phrase, or number.

The periodic table organizes the elements into vertical 1 1. _____ and horizontal 2 in order of increasing 3. The table is 2. _____ constructed so that elements that have similar chemical properties 3. _____ are in the same 4. The elements in Groups 1A through 7A are 4. _____ called the 5. The 6 make up Group 0. The elements in 5. _____ Groups 2A and 3A are interrupted in periods 4 and 5 by the 7 6. _____ and in the periods 6 and 7 by the 8. The Group 1A elements 7. _____ are called the 9, and the Group 2A elements are called the 8. _____ 10. The nonmetals of 11 are called the halogens. 9. _____ Elements with properties that are intermediate between those of 10. _____ metals and nonmetals are called 12. 11. _____ 12. _____

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