

ChemThink: Behavior of Gases

Name _____ Per _____

1. What does the speed of atoms depend on? Describe the relationship between speed and each of these factors.

temperature: \uparrow temp = \uparrow speed
type of atom: smaller/lighter = faster

2. In a sample of one type of atom, do all the atoms move at the same speed? Why?

No, speed is related to each individual atoms energy

3. What is the equation for Pressure and what are the units for pressure?

$$\text{Pressure} = \frac{\text{Force}}{\text{Area}} \quad \text{psi}$$

4. When atoms are moving in a closed container what is creating the force on the sides of the container?

impacts (collisions) of atoms w/ sides of container

5. Explain how a change in temperature causes pressure to change and include an equation.

more collisions results in higher pressure

↓
b/c molecules moving faster

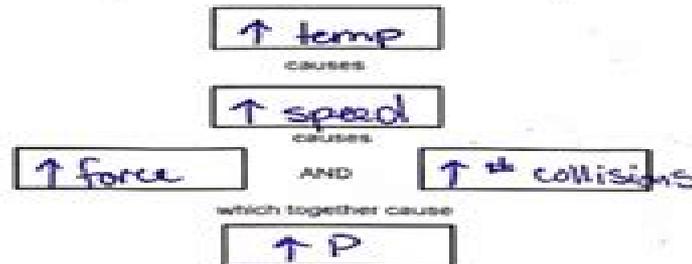
$$\frac{P}{T} = \text{constant}$$

6. Draw the graph you observed of pressure vs. temperature. Is this a direct relationship or an inverse relationship?



direct

7. Fill the cause/effect relationship in the diagram of boxes you observed.



- Explain how the number of atoms present causes pressure to change and include an equation?

more atoms = more collisions, which increases pressure

$$\frac{P}{n} = \text{constant}$$

Behavior Of Gases Chemthink Answers

JR Anderson



Behavior Of Gases Chemthink Answers:

Chemistry and Our Universe , In the first of two lectures on the properties of gases review the basic equations that describe their behavior Learn the history of Boyle s law Gay Lussac s law Charles s law and Avogadro s law Then use these four expressions to derive the celebrated ideal gas law **The Dynamical Theory of Gases (1916)** J. H. Jeans,2008-06-01 This scarce antiquarian book is a facsimile reprint of the original Due to its age it may contain imperfections such as marks notations marginalia and flawed pages Because we believe this work is culturally important we have made it available as part of our commitment for protecting preserving and promoting the world s literature in affordable high quality modern editions that are true to the original work

This book delves into Behavior Of Gases Chemthink Answers. Behavior Of Gases Chemthink Answers is an essential topic that must be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Behavior Of Gases Chemthink Answers, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Behavior Of Gases Chemthink Answers
 - Chapter 2: Essential Elements of Behavior Of Gases Chemthink Answers
 - Chapter 3: Behavior Of Gases Chemthink Answers in Everyday Life
 - Chapter 4: Behavior Of Gases Chemthink Answers in Specific Contexts
 - Chapter 5: Conclusion
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 3. In chapter 2, this book will delve into the foundational concepts of Behavior Of Gases Chemthink Answers. The second chapter will elucidate the essential principles that must be understood to grasp Behavior Of Gases Chemthink Answers in its entirety.
 4. In chapter 3, the author will examine the practical applications of Behavior Of Gases Chemthink Answers in daily life. The third chapter will showcase real-world examples of how Behavior Of Gases Chemthink Answers can be effectively utilized in everyday scenarios.
 5. In chapter 4, the author will scrutinize the relevance of Behavior Of Gases Chemthink Answers in specific contexts. The fourth chapter will explore how Behavior Of Gases Chemthink Answers is applied in specialized fields, such as education, business, and technology.
 6. In chapter 5, this book will draw a conclusion about Behavior Of Gases Chemthink Answers. The final chapter will summarize the key points that have been discussed throughout the book.
- This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Behavior Of Gases Chemthink Answers.

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