

ACID & BASE

- Provide optimal cell function
 - Cells secrete **acid** and **bases** to maintain proper pH for enzymes to work.
- They both provide homeostatic regulation of pH of the body's ECF (extracellular fluid)
 - The Carbonic Acid-Bicarb Buffer System = pH homeostasis of blood

ACID

CO₂ is regulated by the lungs + contributes to acid

- found in the stomach + urine + kidneys *
 - ex: acid reflux, BPH, kidney failure

BASE

Bicarb HCO₃ = form of carbon dioxide in our blood

- found in intestines *
 - Regulated, excreted + reabsorbed in kidneys *
 - Ex C-diff → bacteria in colon

2 FUNCTIONS

Metabolic + Respiratory

METABOLIC

primary change is the concentration of bicarbonate.

METABOLIC ACIDOSIS

bicarb ↓ r/t the amount of acid in the body

- Increased Hydrogen ions
 - CAUSES: kidney failure, DKA, lactic acidosis (shock sepsis), severe diarrhea
 - put pt. on bicarb drip to increase levels
- Respiratory compensation
 - Increased RR to rid of acid

METABOLIC ALKALOSIS

excess bicarb r/t the amount of acid in the body

- Decreased Hydrogen ions
 - CAUSES: loss of gastric fluids, antacid overdose
 - ex: too much NGT suctioning, hypokalemia, vomiting

RESPIRATORY

primary change is the concentration of carbonic acid

RESPIRATORY ACIDOSIS

CO₂ is retained + inadequate O₂ exchange → increases amount of carbonic acid in the body

- CAUSES: Hypoventilation
 - ex: Asthma, COPD, Pneumonia
- Decreased RR
 - Sleep apnea, drug overdose, head injuries

Compensation?

- Increasing RR

RESPIRATORY ALKALOSIS

Too much CO₂ is lost + carbonic acid levels fall

- Pt has increased RR
 - ex: hyperventilation due to anxiety attack

Compensation?

- slow + shallow breaths

C Base Study Guide

J Rink



C Base Study Guide:

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candidates identify focus on and master the specific topics that need the most work Many of the practice questions are taken from previous exams and care was taken to ensure that they cover all the information candidates need to master in order to pass the new computerized Uniform CPA Examination

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