Basic Transport Phenomena in Biomedical Engineering 4th Edition Fournier Solutions Manual

Chapter 1 Solutions

Problem 1.1. Use the conversion factors

$$\frac{8.314\,\text{J}}{\text{mol K}} \times \frac{0.23901\,\text{cal}}{\text{J}} = 1.987\,\frac{\text{cal}}{\text{mol K}}$$

Problem 1.2. Use Equation 1.2 to convert the temperature to degrees Celsius

$$t\,^\circ\text{C} = \frac{5}{9}\left(t\,^\circ\text{F} - 32\right) = \frac{5}{9}\left(98.6 - 32\right) = 37\,^\circ\text{C}$$

Now, use Equation 1.3 to convert the temperature from degrees Celsius to Kelvin

$$T\,\text{K} = t\,^\circ\text{C} + 273.15 = 37 + 273.15 = 310.15\,\text{K}$$

Problem 1.3. Use the conversion factors

$$2\times10^{4}\,\mu\text{dynes} \times \frac{\text{dynes}}{10^{6}\,\mu\text{dynes}} \times \frac{\text{N}}{105\,\text{dynes}} \times \frac{\text{kN}}{10^{3}\,\text{N}} = 2\times10^{-10}\,\text{kN}$$

Problem 1.4. Use the conversion factors

$$\frac{27\,\mu\text{m}}{\text{hr}} \times \frac{10^{-6}\,\text{m}}{\mu\text{m}} \times \frac{0.0006214\,\text{miles}}{\text{m}} \times \frac{\text{hr}}{60\,\text{min}} \times \frac{\text{min}}{60\,\text{sec}} = 4.66\times10^{-12}\,\frac{\text{miles}}{\text{sec}}$$

Problem 1.5. Use the conversion factors

$$48\,\text{MW} \times \frac{10^{6}\,\text{W}}{\text{MW}} \times \frac{\text{kg m}^2}{\text{sec}^3\,\text{W}} \times \frac{10^{3}\,\text{g}}{\text{kg}} \times \frac{100\,\text{cg}}{\text{g}} \times \frac{3600^{3}\,\text{sec}^3}{\text{hr}^3} \times \frac{100^{2}\,\text{cm}^2}{\text{m}^2} = 2.24\times10^{27}\,\frac{\text{cgcm}^2}{\text{hr}^3}$$

$$48\,\text{MW} \times \frac{10^{6}\,\text{W}}{\text{MW}} \times \frac{\text{J}}{\text{sec W}} \times \frac{\text{kJ}}{\cancel{10^{3}\,\text{J}}} \times \frac{60\,\text{sec}}{\cancel{\text{min}}} = 2.88\times10^{6}\,\frac{\text{kJ}}{\cancel{\text{min}}}$$

Problem 1.6. Use the conversion factors to get both values into units of mm per second and then compare the two values for the higher value.

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Liying Dong

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