

Predicting Chemical Equations Exercise Single and Double Replacement

Predict the products in each of the following chemical equations. Balance each of the following. If an equation is already balanced, indicate so. Classify each one as single replacement or double replacement.

- SR 1. $2\text{Na}_{(s)} + 2\text{H}_2\text{O}_{(l)} \rightarrow 2\text{NaOH} + \text{H}_2$
- DR 2. $2\text{NaCl}_{(s)} + \text{H}_2\text{SO}_4_{(aq)} \rightarrow \text{Na}_2\text{SO}_4 + 2\text{HCl}$
- SR 3. $2\text{Al}_{(s)} + \text{Fe}_2\text{O}_3_{(s)} \rightarrow \text{Al}_2\text{O}_3 + 2\text{Fe}$
- DR 4. $\text{Ca}(\text{OH})_2_{(aq)} + \text{Mg}(\text{HCO}_3)_2_{(aq)} \rightarrow \text{Ca}(\text{HCO}_3)_2 + \text{Mg}(\text{OH})_2$
- SR 5. $3\text{K}_{(s)} + \text{AlCl}_3_{(s)} \rightarrow 3\text{KCl} + \text{Al}$
- SR 6. $\text{Cu}_{(s)} + \text{Ag}_2\text{SO}_4_{(aq)} \rightarrow \text{CuSO}_4 + 2\text{Ag}$
- DR 7. $3\text{H}_2\text{SO}_4_{(aq)} + \text{Ca}_3(\text{PO}_4)_2_{(s)} \rightarrow 2\text{H}_3\text{PO}_4 + 3\text{CaSO}_4$
- SR 8. $\text{Cl}_2_{(g)} + \text{MgBr}_2_{(aq)} \rightarrow \text{MgCl}_2 + \text{Br}_2$
- DR 9. $\text{H}_2\text{S}_{(g)} + \text{PbCrO}_4_{(s)} \rightarrow \text{H}_2\text{CrO}_4 + \text{PbS}$
- SR 10. $\text{H}_2\text{S}_{(g)} + 2\text{Ag}_{(s)} \rightarrow \text{Ag}_2\text{S} + \text{H}_2$
- SR 11. $\text{Cl}_2_{(g)} + 2\text{KI}_{(aq)} \rightarrow 2\text{KCl} + \text{I}_2$
- SR 12. $\text{Cu}_{(s)} + 2\text{AgNO}_3_{(aq)} \rightarrow \text{Cu}(\text{NO}_3)_2 + 2\text{Ag}$
- SR 13. $\text{Zn}_{(s)} + 2\text{HCl}_{(aq)} \rightarrow \text{ZnCl}_2 + \text{H}_2$
- SR 14. $\text{Ca}_{(s)} + \text{H}_2\text{SO}_4_{(aq)} \rightarrow \text{CaSO}_4 + \text{H}_2$
- DR 15. $\text{HCl}_{(aq)} + \text{KOH}_{(aq)} \rightarrow \text{H}_2\text{O} + \text{KCl}$
- SR 16. $\text{F}_2_{(g)} + 2\text{KCl}_{(aq)} \rightarrow 2\text{KF} + \text{Cl}_2$
- SR 17. $\text{Zn}_{(s)} + \text{CuSO}_4_{(aq)} \rightarrow \text{ZnSO}_4 + \text{Cu}$
- DR 18. $\text{CaCO}_3_{(s)} + 2\text{HCl}_{(aq)} \rightarrow \text{CaCl}_2 + \text{H}_2\text{CO}_3$
- DR 19. $3\text{NaOH}_{(aq)} + \text{H}_3\text{PO}_4_{(aq)} \rightarrow \text{Na}_3\text{PO}_4 + 3\text{H}_2\text{O}$

10 2 Double Replacement Lab Answers

J Ma



10 2 Double Replacement Lab Answers:

This book delves into 10 2 Double Replacement Lab Answers. 10 2 Double Replacement Lab Answers is a crucial topic that needs to be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into 10 2 Double Replacement Lab Answers, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:
 - Chapter 1: Introduction to 10 2 Double Replacement Lab Answers
 - Chapter 2: Essential Elements of 10 2 Double Replacement Lab Answers
 - Chapter 3: 10 2 Double Replacement Lab Answers in Everyday Life
 - Chapter 4: 10 2 Double Replacement Lab Answers in Specific Contexts
 - Chapter 5: Conclusion
2. In chapter 1, the author will provide an overview of 10 2 Double Replacement Lab Answers. This chapter will explore what 10 2 Double Replacement Lab Answers is, why 10 2 Double Replacement Lab Answers is vital, and how to effectively learn about 10 2 Double Replacement Lab Answers.
3. In chapter 2, the author will delve into the foundational concepts of 10 2 Double Replacement Lab Answers. This chapter will elucidate the essential principles that need to be understood to grasp 10 2 Double Replacement Lab Answers in its entirety.
4. In chapter 3, the author will examine the practical applications of 10 2 Double Replacement Lab Answers in daily life. This chapter will showcase real-world examples of how 10 2 Double Replacement Lab Answers can be effectively utilized in everyday scenarios.
5. In chapter 4, this book will scrutinize the relevance of 10 2 Double Replacement Lab Answers in specific contexts. The fourth chapter will explore how 10 2 Double Replacement Lab Answers is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, this book will draw a conclusion about 10 2 Double Replacement Lab Answers. The final chapter will summarize the key points that have been discussed throughout the book.

The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of 10 2 Double Replacement Lab Answers.

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web secondary school certificate high school examination of the aligarh muslim university or an examination recognized by the aligarh muslim university program objective ssc arts this course aims at understanding the basic concepts of arts social science and for developing strong analytical and reasoning skills to undertake

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web saiyyid hamid senior secondary school boys is a maintained institution of aligarh muslim university and is affiliated to aligarh muslim university board of secondary and senior secondary education aligarh

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web the school was given the new name of s t high school in 1966 after syedna tahir saifuddin then the chancellor of the

aligarh muslim university at present the school is known as s t s school the school has also been imparting education to the students of xi and xii since 1996 1997 for students from outside aligarh the school maintains four

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web established by sir syed ahmad khan as muhammadan anglo oriental collegiate school in 1875 it later evolved as aligarh muslim university and is one of the five senior secondary schools run by the university the school aims primarily at the education of the muslim community