

**QUESTION 1**

- a) Calculate the wavelength of  $\gamma$  rays  $\lambda = 0.00124 \text{ nm}$
- b) Calculate the momentum of  $\gamma$  rays  $p = \frac{h}{\lambda} = \frac{6.626 \times 10^{-34} \text{ J}\cdot\text{s}}{0.00124 \times 10^{-9} \text{ m}} = 5.34 \times 10^{-22} \text{ kg}\cdot\text{m/s}$
- c) Calculate the energy of  $\gamma$  rays  $E = \frac{hc}{\lambda} = \frac{6.626 \times 10^{-34} \text{ J}\cdot\text{s} \times 3 \times 10^8 \text{ m/s}}{0.00124 \times 10^{-9} \text{ m}} = 1.60 \times 10^{-16} \text{ J}$



**QUESTION 2**

- a) Calculate the energy of  $\gamma$  rays  $E = 1.60 \times 10^{-16} \text{ J}$
- b) Calculate the momentum of  $\gamma$  rays  $p = 5.34 \times 10^{-22} \text{ kg}\cdot\text{m/s}$
- c) Calculate the wavelength of  $\gamma$  rays  $\lambda = 0.00124 \text{ nm}$
- d) Calculate the frequency of  $\gamma$  rays  $f = \frac{c}{\lambda} = \frac{3 \times 10^8 \text{ m/s}}{0.00124 \times 10^{-9} \text{ m}} = 2.42 \times 10^{17} \text{ Hz}$



**QUESTION 3**

Calculate the energy of  $\gamma$  rays from the following transitions between energy levels:



Calculate the energy of  $\gamma$  rays from the following transitions between energy levels:



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