

Chemical Reactions

Intro. to Chemical Reactions

Name: _____
 Date: _____
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Electrolysis Reaction

1. Write the chemical formula following compounds:

a) hydrogen gas: H_2 b) sodium phosphate: Na_3PO_4
 c) magnesium phosphate: $Mg_3(PO_4)_2$ d) sodium nitrate: $NaNO_3$

2. Write the correct formula following compounds:

a) $Ca(NO_3)_2$: calcium nitrate b) $PbCl_2$: lead(II) chloride
 c) $CaSO_4$: calcium sulfate d) NaF : sodium fluoride

Chemical Equations: What they look like...

Here is what a chemical reaction looks like when it is written:



There are examples that provide hints to describe what happens during:

Chemical equation reacts with **reactants** and yields **products** and **physical states**

There are hints to solve stoichiometric problems: the numbers and the physical states. Here are a few problems:

Critical Thinking Questions

1. Write the following reaction, with the correct state symbols.



Answer: $2NaCl(aq) + AgNO_3(aq) \rightarrow NaNO_3(aq) + AgCl(s)$

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