ChemThink: Behavior of Gases

Name_____ Per_____

1. What does the speed of atoms depend on? Describe the relationship between speed and each of these factors.

   temperature: ↑ temp = ↑ speed
   type of atom: smaller / lighter = faster

2. In a sample of one type of atom, do all the atoms move at the same speed? Why?

   No, speed is related to each individual atoms energy

3. What is the equation for Pressure and what are the units for pressure?

   $\text{Pressure} = \frac{\text{Force}}{\text{Area}}$ PSi

4. When atoms are moving in a closed container what is creating the force on the sides of the container?

   impacts (collisions) of atoms w/ sides of container

5. **Explain** how a change in temperature causes pressure to change and include an equation.

   more collisions results in higher pressure
   b/c molecules moving faster $\frac{P}{T} = \text{constant}$

6. Draw the graph you observed of pressure vs. temperature. Is this a direct relationship or an inverse relationship?

   direct

7. Fill the cause/effect relationship in the diagram of boxes you observed.

   ↑ temp
   causes
   ↑ speed
   causes
   ↑ force AND ↑ # collisions
   which together cause
   ↑ P

Explain how the number of atoms present causes pressure to change and include an equation?

more atoms = more collisions, which increases pressure

$\frac{P}{n} = \text{constant}$

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Chemthink Lab Answers

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Table of Contents Chemthink Lab Answers

1. Understanding the eBook Chemthink Lab Answers
    - The Rise of Digital Reading Chemthink Lab Answers
    - Advantages of eBooks Over Traditional Books
2. Identifying Chemthink Lab Answers
    - Exploring Different Genres
    - Considering Fiction vs. Non-Fiction
    - Determining Your Reading Goals
3. Choosing the Right eBook Platform
    - Popular eBook Platforms
    - Features to Look for in an Chemthink Lab Answers
    - User-Friendly Interface
4. Exploring eBook Recommendations from Chemthink Lab Answers
    - Personalized Recommendations
    - Chemthink Lab Answers User Reviews and Ratings
    - Chemthink Lab Answers and Bestseller Lists
5. Accessing Chemthink Lab Answers Free and Paid eBooks
    - Chemthink Lab Answers Public Domain eBooks
    - Chemthink Lab Answers eBook Subscription Services

    - Chemthink Lab Answers Budget-Friendly Options
6. Navigating Chemthink Lab Answers eBook Formats
    - ePub, PDF, MOBI, and More
    - Chemthink Lab Answers Compatibility with Devices
    - Chemthink Lab Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Chemthink Lab Answers
    - Highlighting and Note-Taking Chemthink Lab Answers
    - Interactive Elements Chemthink Lab Answers
8. Staying Engaged with Chemthink Lab Answers
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Chemthink Lab Answers
9. Balancing eBooks and Physical Books Chemthink Lab Answers
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Chemthink Lab Answers
10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
11. Cultivating a Reading Routine Chemthink Lab Answers
    - Setting Reading Goals Chemthink Lab Answers
    - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemthink Lab Answers
    - Fact-Checking eBook Content of Chemthink Lab Answers
    - Distinguishing Credible Sources
13. Promoting Lifelong Learning
    - Utilizing eBooks for Skill Development
    - Exploring Educational eBooks
14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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