ChemThink: Ionic Bonding

Ionic Bonds are formed between _____ and _____ and involve the _____ of ions.

Charges that are alike will _____ each other, while opposite charges will _____ and _____ together.

In order to build an ionic compound, you must have both a _____ ion and a _____ ion.

The positive ion will be formed from an atom that has a tendency to _____ electrons and will usually be a _____.

A negative ion will be formed from an atom that has a tendency to _____ electrons and will usually be a _____.

Conclusion: _____ and _____ can bond together by forming an ionic bond.

When multiple ion pairs are attracted to each other it forms an ionic _____

In ionic compounds the formula tells us the _____ between _____ in the crystal.

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A Loxley

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