

# Voltaic Cells

Label the parts for the following cell:



By connecting to the voltmeter, describe a spontaneous redox reaction. (10 marks)

- anode
- cathode
- anion
- cation
- electron movement flow
- direction of ion movement
- half-cell reduction equation
- Zn
- Write overall cell equation with phases.  $\text{Zn(s)} + \text{Cu}^{2+}(\text{aq}) \rightarrow \text{Zn}^{2+}(\text{aq}) + \text{Cu(s)}$  (10 marks)

\_\_\_\_\_

# Voltaic Cells Study Guide

**Kenneth L. Henold**



## **Voltaic Cells Study Guide:**

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Frentrup,2002-06 Contains a brief overview of every chapter review of skills self tests and the answers and detailed solutions to all odd numbered end of chapter problems in the text book

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in a bucket or bowl the spruce adelyn duchala

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