

## Student Exploration: Covalent Bonds

<p><b>Activity B:</b> <b>Building molecules</b></p>	<p><u>Get the Gizmo ready:</u></p> <ul style="list-style-type: none"> <li>• Click <b>Reset</b>.</li> <li>• Turn off <b>Show Lewis diagram</b>.</li> <li>• Select <b>Oxygen</b>.</li> </ul>	
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**Question:** How do atoms share more than one pair of electrons?

1. **Observe:** Like fluorine and most other elements, oxygen atoms are most stable with a full complement of eight valence electrons.

A. How many valence electrons does each oxygen atom have now? 6

B. How many more electrons does each oxygen atom need to be stable? 2

2. **Form a bond:** Drag electrons back and forth until the molecule of oxygen ( $O_2$ ) is stable. Click **Check** to confirm your molecule is stable. Take a snapshot and paste the image into your document (don't forget to label it " $O_2$ ").

How many *pairs* of shared electrons are there in a stable molecule of oxygen? 1

3. **Draw a diagram:** Draw a Lewis diagram of the oxygen molecule in the space below at left. To check your work, turn on **Show Lewis diagram**. Draw the correct diagram on the right.

Practice diagram:  $:\ddot{O} = \ddot{O}:$

Actual:  $:\ddot{O} = \ddot{O}:$

4. **Practice:** Create covalent bonds and stable molecules for the remaining substances. Take a snapshot of each completed molecule and add it to your document. Draw Lewis diagrams for each one. (As above, draw the diagram on your own before checking your work.)

Nitrogen



Ammonia



Chlorine



Methane



Water



Carbon dioxide



Formaldehyde



(Activity B continued on next page)

# Covalent Bond Explore Learning Answer Key

**Camilla Rothe**



## Covalent Bond Explore Learning Answer Key:

**Chemistry** Trace Jordan, Neville R. Kallenbach, 2017 *Chemistry The Molecules of Life* offers chemical insights within the context of health pharmaceuticals and the function of biological molecules The contextualized presentation of topics gives students a broad introduction to chemistry and helps them to see the relevance of chemistry to their personal lives

Chemical Interactions, 2005 Favorite Demonstrations for College Science Brian Shmaefsky, 2004 The book is an all in one compilation of 36 popular classroom demonstrations published since 1993 in the Favorite Demonstration column of NSTA's *Journal of College Science Teaching* The collection begins with a chapter on safety The Rules of Research From there chapters emphasize conveying scientific principles while making them memorable **Principles of Modern Chemistry**

David W. Oxtoby, H. Pat Gillis, Laurie J. Butler, 2016-01-01 Long considered the standard for honors and high level mainstream general chemistry courses *PRINCIPLES OF MODERN CHEMISTRY* continues to set the standard as the most modern rigorous and chemically and mathematically accurate text on the market This authoritative text features an atoms first approach and thoroughly revised chapters on Quantum Mechanics and Molecular Structure Chapter 6 Electrochemistry Chapter 17 and Molecular Spectroscopy and Photochemistry Chapter 20 In addition the text utilizes mathematically accurate and artistic atomic and molecular orbital art and is student friendly without compromising its rigor End of chapter study aids focus on only the most important key objectives equations and concepts making it easier for students to locate chapter content while applications to a wide range of disciplines such as biology chemical engineering biochemistry and medicine deepen students understanding of the relevance of chemistry beyond the classroom *Introduction to Zeolite Molecular Sieves* Jiri Cejka, Herman van Bekkum, A. Corma, F. Schueth, 2007-10-02 *Introduction to Zeolite Molecular Sieves* 3rd Edition presents a collection of the most important results and ideas in the field of molecular sieve chemistry and technology the most important experimental techniques related to the research activities in molecular sieves and identifies new areas of molecular sieve chemistry Chapters start at a reasonably simple entry level but also covers the present state of the art in the field Topics covered include structure synthesis characterization ion exchange adsorption diffusion separations and natural zeolites 6 years since the last edition this book brings together the rapid development within the field of molecular sieve chemistry and applications Accessible to newcomers to the field also containing valuable information for experienced researchers 27 chapters written by renowned scientists in their field including updates on some 2nd edition chapters

*Investigating Chemistry* Matthew Johll, 2008-12-22 In its new second edition *Investigating Chemistry A Forensic Science Perspective* remains the only book that uses the inherently fascinating topics of crime and criminal investigations as a context for teaching the fundamental chemical concepts most often covered in an introductory nonmajors course Covering all the standard topics Matthew Johll capitalizes on the surge of interest in the scientific investigation of crime as sparked by CSI and other television shows bringing together the theme of forensic science and the fundamentals of chemistry in ways

that are effective and accessible for students This edition features refined explanations of the chemical concepts which are the core of the book as well as a more thoroughly integrated forensic theme updated features and an expanded media supplements package

**Exploring Mechanistic Reasoning in Science Education** Nicole Graulich, Michal Haskel Ittah, Rayendra Wahyu Bachtiar, 2026-02-10 This edited volume provides a comprehensive exploration of the diverse dimensions of mechanistic reasoning within science education inviting to understand its significance development and practical applications across diverse scientific disciplines Through its four sections it covers mechanistic reasoning through diverse lenses such as research studies instructional strategies the nature of explanations and its overall impact on scientific literacy It combines theoretical discussions with empirical evidence and practical applications offering a multifaceted perspective on mechanistic reasoning This book is an important contribution to science educators working on mechanistic reasoning and related fields such as explanations system thinking modeling and argumentation Additionally its practical recommendations regarding approaches for evaluating and promoting mechanistic reasoning make this a volume of interest for curriculum designers and teachers

**Educational Film/video Locator of the Consortium of University Film Centers and R.R. Bowker** Consortium of University Film Centers, 1986

**Peer-Led Team Learning** David Gosser, Victor S. Stozak, Mark S. Cracolice, 2006 The Workshop Program is a collaboration of teachers learning specialists and students that develops and implements a Peer Led Team Learning model of teaching science Students who have done well in the course are selected and trained for leadership roles These peer leaders guide the work of teams of 6-8 students to solve carefully structured problems in weekly Workshop sessions

**Peer-led Team Learning** David K. Gosser, Victor S. Stozak, Mark S. Cracolice, 2001 Workbook developed from the Workshop Chemistry Project which explored developed and applied the concept of peer led team learning in problem solving workshops in introductory chemistry courses

*An Introduction to Physical Science* James T. Shipman, Jerry D. Wilson, Aaron W. Todd, 2002-07-08 This text presents a survey of the physical sciences physics chemistry astronomy meteorology and geology for non science majors Topics are treated both descriptively and quantitatively providing flexibility for instructors who wish to emphasize a highly descriptive approach a highly quantitative approach or any spectrum in between

**Science Scope**, 2000

**Holt Chemistry** Ralph Thomas Myers, 2004

**The Software Encyclopedia 2001**, 2001

Toxicology Research Projects Directory, 1980 An indexed directory of current research project abstracts in toxicology and related fields

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