



The War for Europe and North Africa

Learning

By the end of this session you should be able to:

- 1. Identify the main causes of the war.
- 2. Explain the main events of the war.
- 3. Describe the main events of the war.
- 4. Explain the main events of the war.
- 5. Describe the main events of the war.
- 6. Explain the main events of the war.

Evaluating

By the end of this session you should be able to:

- 1. Explain the main events of the war.

2. Explain the main events of the war.

3. Explain the main events of the war.

4. Explain the main events of the war.

5. Explain the main events of the war.

Chapter 2section Reteaching Activity War In Europe

IM Harris



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 $2 \text{ CO}_2(\text{g}) + \text{H}_2\text{O}(\text{g}) \rightarrow \text{C}_2\text{H}_2(\text{g}) + \frac{5}{2} \text{O}_2(\text{g})$ $\Delta h = 94.5 \text{ kJ}$
 $\text{H}_2\text{O}(\text{g}) \rightarrow \text{H}_2(\text{g}) + \frac{1}{2} \text{O}_2(\text{g})$ $\Delta h = 71.2 \text{ kJ}$
 $\text{C}_2\text{H}_6(\text{g}) + \frac{7}{2} \text{O}_2(\text{g}) \rightarrow 2 \text{ CO}_2(\text{g}) + 3 \text{ H}_2\text{O}(\text{g})$ $\Delta h = 283 \text{ kJ}$
 $\text{C}_2\text{H}_6(\text{g})$

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 $\text{C}_2\text{H}_4(\text{g}) + 3 \text{ O}_2(\text{g}) \rightarrow 2 \text{ CO}_2(\text{g}) + 2 \text{ H}_2\text{O}(\text{l})$ $\Delta h = 1411 \text{ kJ/mole}$
 $\text{C}_2\text{H}_6(\text{g}) + \frac{7}{2} \text{ O}_2(\text{g}) \rightarrow 2 \text{ CO}_2(\text{g}) + 3 \text{ H}_2\text{O}(\text{l})$ $\Delta h = 1560 \text{ kJ/mole}$
 $\text{H}_2(\text{g}) + \frac{1}{2} \text{ O}_2(\text{g}) \rightarrow \text{H}_2\text{O}(\text{l})$ $\Delta h = 285 \text{ kJ/mole}$

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web thus for reaction 9 2 in which 1 mole of h_2o and 1 mole of oh^- combine to form 2 moles of h_2o $h = 55.8 \text{ kJ}$ as shown note that the enthalpy change $h_{products} - h_{reactants}$ is positive if heat is absorbed that is if $h_{products}$

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