Name _____ Date _____ Class _____

CHAPTER 6 STUDY GUIDE FOR CONTENT MASTERY

Section 6.1 *continued*

In your textbook, read about the modern periodic table.

Use the information in the box on the left to fill in the periodic table to complete the table on the right.

11
16
Nitrogen
14.007
$[Ne]3s^2 3p^2$

Atomic Mass	**12.**
Atomic Number	**14.**
Electron Configuration	**15.**
Chemical Name	**16.**
Chemical Symbol	**17.**

For each item in Column A, write the letter of the matching item in Column B.

Column A	Column B
_____ **18.** A column on the periodic table	**a.** metals
_____ **19.** A row on the periodic table	**b.** group
_____ **20.** Group A elements	**c.** period
_____ **21.** Elements that are shiny and conduct electricity	**d.** representative elements
_____ **22.** Group B elements	**e.** transition elements

In the space on the left, write true if the statement is true, if the statement is false, change the indicated word or phrase to make it true.

_____ **23.** There are two main classifications of elements.

_____ **24.** Most than three-fourths of the elements in the periodic table are nonmetals.

_____ **25.** Group 1A elements occupy two hydrogens and known as the actual metals.

_____ **26.** Group 2A elements are the alkaline earth metals.

_____ **27.** Group 7A elements are highly reactive nonmetals known as halogens.

_____ **28.** Group 8A elements are very nonreactive elements known as transition metals.

_____ **29.** Metallically have properties of both metals and more transition metals.

Chapter 3 Study Guide For Content Mastery Answer Key

Wenbin Ji

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