

CHAPTER 8 STUDY GUIDE

Covalent Bonding

Section 8.1 The Covalent Bond

In your textbook, read about the nature of covalent bonds.

Use each of the terms below just once to complete the passage.

covalent bond	molecule	sigma bond	exothermic	pi bond
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When sharing of electrons occurs, the attachment between atoms that results is called a(n) **(1)** _____. When such an attachment is formed, bond dissociation energy is released, and the process is **(2)** _____. When two or more atoms bond by means of electron sharing, the resulting particle is called a(n) **(3)** _____. If the electrons shared are centered between the two atoms, the attachment is called a(n) **(4)** _____. If the sharing involves the overlap of parallel orbitals, the attachment is called a(n) **(5)** _____.

In your textbook, read about single and multiple bonds and bond strength.

Circle the letter of the choice that best completes the statement or answers the question.

- In what form do elements such as hydrogen, nitrogen, and oxygen normally occur?
 - as single atoms
 - as molecules containing two atoms
 - as molecules containing three atoms
 - as molecules containing four atoms
- How many electrons are shared in a double covalent bond?
 - none
 - one
 - two
 - four
- Bond length is the distance between
 - two molecules of the same substance.
 - the electrons in two attached atoms.
 - the nuclei of two attached atoms.
 - the orbitals of two attached atoms.
- Which of the following relationships relating to bond length is generally correct?
 - the shorter the bond, the stronger the bond
 - the shorter the bond, the weaker the bond
 - the shorter the bond, the fewer the electrons in it
 - the shorter the bond, the lower the bond dissociation energy

Chapter 9 Covalent Bonding Study Guide Answers

Peipei Pang



Chapter 9 Covalent Bonding Study Guide Answers:

Whispering the Techniques of Language: An Emotional Quest through **Chapter 9 Covalent Bonding Study Guide Answers**

In a digitally-driven earth wherever screens reign supreme and instant connection drowns out the subtleties of language, the profound secrets and emotional subtleties hidden within words frequently go unheard. However, situated within the pages of **Chapter 9 Covalent Bonding Study Guide Answers** a fascinating fictional prize blinking with organic emotions, lies an extraordinary quest waiting to be undertaken. Written by an experienced wordsmith, this charming opus attracts visitors on an introspective trip, gently unraveling the veiled truths and profound impact resonating within the fabric of each and every word. Within the psychological depths of this moving evaluation, we shall embark upon a genuine exploration of the book's primary styles, dissect its fascinating writing design, and yield to the strong resonance it evokes deep within the recesses of readers' hearts.

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