

CHAPTER 8 REVIEW

Chemical Equations and Reactions**MIXED REVIEW****SHORT ANSWER** Answer the following questions in the space provided. (1 point)

- _____ A balanced chemical equation represents all the following *except*
 - experimentally established facts.
 - the mechanism by which reactants combine to form products.
 - identities of reactants and products in a chemical reaction.
 - relative quantities of reactants and products in a chemical reaction.
- _____ According to the law of conservation of mass, the total mass of the reacting substances is
 - always more than the total mass of the products.
 - always less than the total mass of the products.
 - sometimes more and sometimes less than the total mass of the products.
 - always equal to the total mass of the products.

3. Predict whether each of the following chemical reactions will occur. For each reaction that will occur, identify the reaction type and complete the chemical equation by writing in the products formed and balancing the final equation. (2 points each)









Chapter Mixed Review Chemistry

S Ashworth



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