

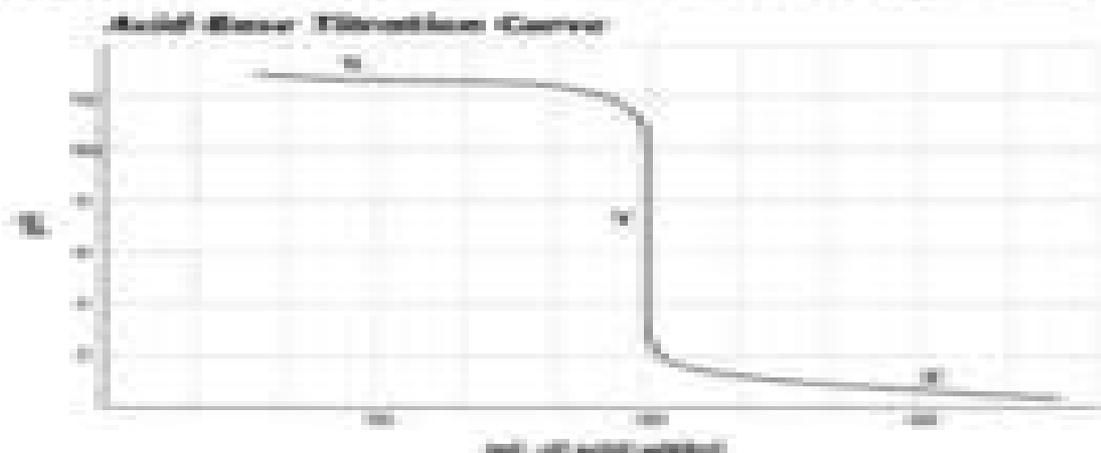
CHAPTER 14 REVIEW

Acid-Base Titration and pH

SECTION 2

SHORT ANSWER Answer the following questions in the space provided.

1. Below is a pH curve from an acid-base titration. The x-axis labels show points A, B, C, and D.



- _____ a. Which point represents the equivalence point?
- _____ b. At which point is there excess acid in the system?
- _____ c. At which point is there excess base in the system?
- _____ d. If the three solutions in (b)-(c) are acid-base conjugates of each other, how would you identify them? Name every member of this conjugate acid-base pair.

PROBLEM Write the answer on the line to the left. Show all your work in the space provided.

1. A 100-mL solution of 0.100 M HCl is mixed with a 100-mL solution of sodium hydroxide to decrease its acidity and its volume. It takes 25 mL of the base to neutralize the HCl in the acid.
- a. Write the balanced molecular equation for this neutralization reaction.

Chapter Review Acid Base Titration Ph Answers

Ying Liu



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