

Experiment 2 - Gravimetric Analysis of a Chloride Salt (G)

Purpose:

Determining quantitative value of chloride in an unknown soluble salt by showing and explaining common methods in gravimetric analysis.

Theory:

- Reaction between silver ion and chloride ion to form silver chloride as an equation:



- AgCl solubility in water and the K_{sp} :

Silver chloride is very insoluble, so we can assume that the above reaction goes to completion.

$$K_{sp} = 1.6 \times 10^{-10}$$

- AgCl is precipitated with an acidic solution to reduce likelihood of acid ions interfering with results. These ions will form co-precipitates with silver in the acid, due to these co-precipitates, the results would be higher.
- To prevent the precipitate from going through the filtered paper, it is heated to coagulate it into a different texture that's easier to work with. Heating the precipitate makes it coagulate, therefore, it is harder for the precipitate to go through the filtered paper.
- Dry silver chloride decomposes into silver and chlorine in the presence of light:



- If photodecomposition takes place in air, results will be lower, since chlorine will be released in the form of gas. If photochemical decomposition occurs in the presence of excess silver ions in aqueous solution, it will result in higher results.

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Geological Survey of Texas



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English in Analytical Chemistry. Communicating about Methods & Techniques. Книга для студента Надежда Зорина, Александра Соболева, 2022-10-19

Laboratory Experiments for General Chemistry Harold R. Hunt, Toby F. Block, 1994

Chemistry in the Laboratory James M. Postma, Julian L. Robert, J. Leland Hollenberg, 2004-03-12 This clearly written class tested manual has long given students hands on experience covering all the essential topics in general chemistry Stand alone experiments provide all the background introduction necessary to work with any general chemistry text This revised edition offers new experiments and expanded information on applications to real world situations

General, Organic, and Biological Chemistry Study Guide and Selected Solutions Karen C. Timberlake, 2001-11 Keyed to the learning goals in the text this guide is designed to promote active learning through a variety of exercises with answers and mastery exams The guide also contains complete solutions to odd numbered problems

Scientific and Technical Aerospace Reports, 1981-10

Experiments in General Chemistry Kenneth W. Whitten, Kenneth D. Gailey, 1981

General Chemistry Hunt, 1976-06-07

1998 Report on Attainment of the California Particulate Matter Standards in the Monterey Bay Region Janet Brennan, 1999

Cooperative Chemistry Lab Manual Cooper, 2005-02 The laboratory course described in the lab manual emphasizes experimental design data analysis and problem solving Inherent in the design is the emphasis on communication skills both written and oral Students work in groups on open ended projects in which they are given an initial scenario and then asked to investigate a problem There are no formalized instructions and students must plan and carry out their own investigations

Annual Report University of Wisconsin--Madison. Center for Dairy Research, 1997

Chemical Investigations for Changing Times C. Alton Hassell, John Hill, 2001

Chemistry Karen Timberlake, 1999

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environmental science and environmental decision making that is essential for finding solutions to environmental problems

Chemical Investigations Nancy Konigsberg Kerner, 1986 **Report of the Commissioner of Education** USA.
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Table of Contents Analysis Of An Unknown Salt Lab Report

1. Understanding the eBook Analysis Of An Unknown Salt Lab Report
 - The Rise of Digital Reading Analysis Of An Unknown Salt Lab Report
 - Advantages of eBooks Over Traditional Books
2. Identifying Analysis Of An Unknown Salt Lab Report
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Analysis Of An Unknown Salt Lab Report
 - User-Friendly Interface

4. Exploring eBook Recommendations from Analysis Of An Unknown Salt Lab Report
 - Personalized Recommendations
 - Analysis Of An Unknown Salt Lab Report User Reviews and Ratings
 - Analysis Of An Unknown Salt Lab Report and Bestseller Lists
5. Accessing Analysis Of An Unknown Salt Lab Report Free and Paid eBooks
 - Analysis Of An Unknown Salt Lab Report Public Domain eBooks
 - Analysis Of An Unknown Salt Lab Report eBook Subscription Services
 - Analysis Of An Unknown Salt Lab Report Budget-Friendly Options
6. Navigating Analysis Of An Unknown Salt Lab Report eBook Formats
 - ePub, PDF, MOBI, and More
 - Analysis Of An Unknown Salt Lab Report Compatibility with Devices
 - Analysis Of An Unknown Salt Lab Report Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Analysis Of An Unknown Salt Lab Report
 - Highlighting and Note-Taking Analysis Of An Unknown Salt Lab Report
 - Interactive Elements Analysis Of An Unknown Salt Lab Report
8. Staying Engaged with Analysis Of An Unknown Salt Lab Report
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Analysis Of An Unknown Salt Lab Report
9. Balancing eBooks and Physical Books Analysis Of An Unknown Salt Lab Report
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Analysis Of An Unknown Salt Lab Report
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Analysis Of An Unknown Salt Lab Report
 - Setting Reading Goals Analysis Of An Unknown Salt Lab Report
 - Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Analysis Of An Unknown Salt Lab Report
 - Fact-Checking eBook Content of Analysis Of An Unknown Salt Lab Report
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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