

Balancing Equations and Predicting Products

2 Part 1: Balancing Equations

Directions: Balance the following chemical equations and identify the type of reaction.

- $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ Type of reaction: Synthesis
- $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
Type of reaction: Acid base
- $\text{Mg} + \text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2$
Type of reaction: Single displacement
- $\text{NaCl} + \text{AgNO}_3 \rightarrow \text{NaNO}_3 + \text{AgCl}$
Type of reaction: Double displacement
- $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$
Type of reaction: Decomposition
- $\text{C}_2\text{H}_6 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
Type of reaction: Combustion

3 Part 2: Predicting Products

Directions: Given the reactants, predict the products of the following chemical reactions.

- $\text{Na} + \text{Cl}_2 \rightarrow ?$
- $\text{Ca}(\text{OH})_2 + \text{H}_2\text{SO}_4 \rightarrow ?$
- $\text{Mg}(\text{NO}_3)_2 + \text{NaOH} \rightarrow ?$
- $\text{HNO}_3 + \text{K} \rightarrow ?$

Balancing And Predicting Products

M Carnoy



Balancing And Predicting Products:

Decoding **Balancing And Predicting Products**: Revealing the Captivating Potential of Verbal Expression

In a period characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its ability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Balancing And Predicting Products**," a mesmerizing literary creation penned by a celebrated wordsmith, readers attempt an enlightening odyssey, unraveling the intricate significance of language and its enduring effect on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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