

Name: _____

Balancing Chemical Equations



Directions: Balance each of the following chemical equations.

- 1) $\underline{\quad} \text{H}_3\text{PO}_4 + \underline{\quad} \text{KOH} \rightarrow \underline{\quad} \text{K}_3\text{PO}_4 + \underline{\quad} \text{H}_2\text{O}$
- 2) $\underline{\quad} \text{K} + \underline{\quad} \text{B}_2\text{O}_3 \rightarrow \underline{\quad} \text{K}_2\text{O} + \underline{\quad} \text{B}$
- 3) $\underline{\quad} \text{HCl} + \underline{\quad} \text{NaOH} \rightarrow \underline{\quad} \text{NaCl} + \underline{\quad} \text{H}_2\text{O}$
- 4) $\underline{\quad} \text{Na} + \underline{\quad} \text{NaNO}_3 \rightarrow \underline{\quad} \text{Na}_2\text{O} + \underline{\quad} \text{N}_2$
- 5) $\underline{\quad} \text{C} + \underline{\quad} \text{S}_8 \rightarrow \underline{\quad} \text{CS}_2$
- 6) $\underline{\quad} \text{Na} + \underline{\quad} \text{O}_2 \rightarrow \underline{\quad} \text{Na}_2\text{O}$
- 7) $\underline{\quad} \text{N}_2 + \underline{\quad} \text{O}_2 \rightarrow \underline{\quad} \text{N}_2\text{O}_5$
- 8) $\underline{\quad} \text{H}_3\text{PO}_4 + \underline{\quad} \text{Mg}(\text{OH})_2 \rightarrow \underline{\quad} \text{Mg}_3(\text{PO}_4)_2 + \underline{\quad} \text{H}_2\text{O}$
- 9) $\underline{\quad} \text{NaOH} + \underline{\quad} \text{H}_2\text{CO}_3 \rightarrow \underline{\quad} \text{Na}_2\text{CO}_3 + \underline{\quad} \text{H}_2\text{O}$
- 10) $\underline{\quad} \text{KOH} + \underline{\quad} \text{HBr} \rightarrow \underline{\quad} \text{KBr} + \underline{\quad} \text{H}_2\text{O}$
- 11) $\underline{\quad} \text{Na} + \underline{\quad} \text{O}_2 \rightarrow \underline{\quad} \text{Na}_2\text{O}$
- 12) $\underline{\quad} \text{Al}(\text{OH})_3 + \underline{\quad} \text{H}_2\text{CO}_3 \rightarrow \underline{\quad} \text{Al}_2(\text{CO}_3)_3 + \underline{\quad} \text{H}_2\text{O}$
- 13) $\underline{\quad} \text{Al} + \underline{\quad} \text{S}_8 \rightarrow \underline{\quad} \text{Al}_2\text{S}_3$
- 14) $\underline{\quad} \text{Cs} + \underline{\quad} \text{N}_2 \rightarrow \underline{\quad} \text{Cs}_3\text{N}$
- 15) $\underline{\quad} \text{Mg} + \underline{\quad} \text{Cl}_2 \rightarrow \underline{\quad} \text{MgCl}_2$
- 16) $\underline{\quad} \text{Rb} + \underline{\quad} \text{RbNO}_3 \rightarrow \underline{\quad} \text{Rb}_2\text{O} + \underline{\quad} \text{N}_2$
- 17) $\underline{\quad} \text{C}_6\text{H}_6 + \underline{\quad} \text{O}_2 \rightarrow \underline{\quad} \text{CO}_2 + \underline{\quad} \text{H}_2\text{O}$
- 18) $\underline{\quad} \text{N}_2 + \underline{\quad} \text{H}_2 \rightarrow \underline{\quad} \text{NH}_3$

Balancing Chemical Equations Kit Answers

Clive Long



Balancing Chemical Equations Kit Answers:

X-kit Fet G11 Phys Science Chemist Clive Long,2010

The Explore-Before-Explain Guidebook for Science

Education Patrick Brown,2025-06-30 This guidebook uses an Explore before Explain instructional sequence to help you facilitate the design of active meaning making lessons in science Author Pat Brown models and breaks down how an Explore before Explain approach ensures students conceptual understandings are constructed primarily on evidence based experiences in the classroom From prompting students to ponder patterns helping them identify cause and effect relationships to focusing on development of their thinking rather than validating ideas you can use Explore before Explain instruction to help your students feel confident in their thinking and become more self directed learners Chapters focus on developing your students conceptual science understanding through the investigation of natural occurrences Content and module examples are provided as well as attention to contemporary standards and safety in science Rather than acting as a prescriptive program however this book adds another element to your curriculum design showing how lessons can and should include critical elements of active sensemaking when designing an Explore before Explain sequence In addition the author shows the benefits of bringing Explore before Explain outside the classroom to create high quality professional and curriculum based learning This resource is ideal for K 12 science teachers as well as building administrators who are looking for a practice oriented and research based approach for their science curriculum As a classroom educator you can use these strategies for leveraging research into hands on minds on activities to promote more robust and equitable learning environments For leaders this resource can be used to develop professional learning experiences for sustained departmental improvement

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