

Name: \_\_\_\_\_



## Balancing Chemical Equations

Directions: Balance each of the following chemical equations.

- 1)  $\_\_\_ \text{C}_{30}\text{H}_{72} + \_\_\_ \text{O}_2 \rightarrow \_\_\_ \text{CO}_2 + \_\_\_ \text{H}_2\text{O}$
- 2)  $\_\_\_ \text{Al}(\text{OH})_3 + \_\_\_ \text{HBr} \rightarrow \_\_\_ \text{AlBr}_3 + \_\_\_ \text{H}_2\text{O}$
- 3)  $\_\_\_ \text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_3 + \_\_\_ \text{O}_2 \rightarrow \_\_\_ \text{CO}_2 + \_\_\_ \text{H}_2\text{O}$
- 4)  $\_\_\_ \text{C}_7\text{H}_8 + \_\_\_ \text{O}_2 \rightarrow \_\_\_ \text{CO}_2 + \_\_\_ \text{H}_2\text{O}$
- 5)  $\_\_\_ \text{Li} + \_\_\_ \text{AlCl}_3 \rightarrow \_\_\_ \text{LiCl} + \_\_\_ \text{Al}$
- 6)  $\_\_\_ \text{C}_3\text{H}_6 + \_\_\_ \text{O}_2 \rightarrow \_\_\_ \text{CO}_2 + \_\_\_ \text{H}_2\text{O}$
- 7)  $\_\_\_ \text{NH}_4\text{OH} + \_\_\_ \text{H}_3\text{PO}_4 \rightarrow \_\_\_ (\text{NH}_4)_3\text{PO}_4 + \_\_\_ \text{H}_2\text{O}$
- 8)  $\_\_\_ \text{Rb} + \_\_\_ \text{P} \rightarrow \_\_\_ \text{Rb}_3\text{P}$
- 9)  $\_\_\_ \text{CH}_4 + \_\_\_ \text{O}_2 \rightarrow \_\_\_ \text{CO}_2 + \_\_\_ \text{H}_2\text{O}$
- 10)  $\_\_\_ \text{Al}(\text{OH})_3 + \_\_\_ \text{H}_2\text{SO}_4 \rightarrow \_\_\_ \text{Al}_2(\text{SO}_4)_3 + \_\_\_ \text{H}_2\text{O}$
- 11)  $\_\_\_ \text{Na} + \_\_\_ \text{Cl}_2 \rightarrow \_\_\_ \text{NaCl}$
- 12)  $\_\_\_ \text{Rb} + \_\_\_ \text{S}_8 \rightarrow \_\_\_ \text{Rb}_2\text{S}$
- 13)  $\_\_\_ \text{H}_3\text{PO}_4 + \_\_\_ \text{Ca}(\text{OH})_2 \rightarrow \_\_\_ \text{Ca}_3(\text{PO}_4)_2 + \_\_\_ \text{H}_2\text{O}$
- 14)  $\_\_\_ \text{NH}_3 + \_\_\_ \text{HCl} \rightarrow \_\_\_ \text{NH}_4\text{Cl}$
- 15)  $\_\_\_ \text{Li} + \_\_\_ \text{H}_2\text{O} \rightarrow \_\_\_ \text{LiOH} + \_\_\_ \text{H}_2$
- 16)  $\_\_\_ \text{Ca}_3(\text{PO}_4)_2 + \_\_\_ \text{SiO}_2 + \_\_\_ \text{C} \rightarrow \_\_\_ \text{CaSiO}_3 + \_\_\_ \text{CO} + \_\_\_ \text{P}$
- 17)  $\_\_\_ \text{NH}_3 + \_\_\_ \text{O}_2 \rightarrow \_\_\_ \text{N}_2 + \_\_\_ \text{H}_2\text{O}$
- 18)  $\_\_\_ \text{FeS}_2 + \_\_\_ \text{O}_2 \rightarrow \_\_\_ \text{Fe}_2\text{O}_3 + \_\_\_ \text{SO}_2$

# Balancing Equations Chemfiesta Answers

**Yan Bai**



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web miss earth 2023 will be the 23rd miss earth pageant to be held at the saigon exhibition and convention center in ho chi minh city vietnam on 22 december 2023 mina sue choi of south korea will crown her successor at the end of

**miss earth indonesia 2023 is cindy inanto pageantcircle com** - Mar 10 2023

web sep 13 2023 wednesday september 13 2023 cindy inanto was crowned miss earth indonesia 2023 during the putri nusantara 2023 final held last july 27 at renaissance nusa dua in bali the 27 year old computer engineer and data scientist succeeded last year's titleholder karina basrewan and will now represent indonesia at miss earth 2023

**miss earth 2022 wikipedia** - Apr 11 2023

web miss earth 2022 official logo miss earth 2022 was the 22nd miss earth pageant held on november 29 2022 at the okada manila in parañaque metro manila philippines the competition returned to the philippines into a standard format for the first time since 2019 after easing travel restrictions surrounding the covid 19 pandemic

miss earth indonesia 2022 - Jan 08 2023

web miss earth indonesia 2022 komodo dragons are a representation of confident old souls they move with purpose and strategy and they are not afraid to go into things alone when everyone leaves them behind indonesia karina basrewan height 164 cm measurements 33 in 26 in 38 in career education tv news anchor and community

**miss earth indonesia 2023 vietnam miss earth indonesia** - Nov 06 2022

web miss earth indonesia 2023 vietnam miss earth indonesia 2023 is cindy inanto she was crowned by miss earth indonesia 2022 karina basrewan and miss earth 2022 mina sue choi by miss earth moments facebook video