

Balancing Chemical Equations

In any chemical reaction, the chemicals that are mixed together are called the reactants. The reactants are written to the left of the arrow. The chemicals that are produced in the reaction are called the products.



In the above reaction we see that one molecule of H_2 reacts with one molecule of O_2 to make one molecule of H_2O . Let's use the following table to summarize the total number of atoms on each side of the equation:

Reactants	Products
2 - H's	2 - H's
2 - O's	1 - O

This equation is not "balanced" at the present time because there are different number of oxygen atoms on each side of the arrow. We can balance the equation however, by adding some coefficients.



Notice what we have done. We have equaled out the number of hydrogen and oxygen atoms on each side of the equation. Let's look at a summary of the atoms present now.

Reactants	Products
4 - H's	4 - H's
2 - O's	2 - O's

Equations such as this one, in which the number of each of the individual atoms are equal on both sides of the equation, are said to be balanced.

One key point that we need to stress when we work on balancing chemical equations is that we only change the coefficients; never the subscripts, in order to achieve equal numbers of atoms on each side.



The 5 is called a coefficient; and coefficients can be changed to balance reactions. Changing a coefficient simply changes the number of molecules present, but does not affect the composition of the molecules.

The 2 & 4 are called subscripts; and subscripts can't be changed in order to balance an equation. Changing any of the subscripts would result in a new compound which we do not want.

Let's take a few minutes and see if we can balance a few practice equations together.



So far, we have balanced our equations by "inspection." In other words simply by a combination of logic and trial-and-error. This is usually the fastest method to use, however, as you will see later in the section on oxidation and reduction, there are systematic methods to use that can help you balance equations.

Let's take a few minutes to learn a few tricks that might help you balance some equations...

Balancing Equations Packet

VM Jensen



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