

BALANCING CHEMICAL EQUATIONS

Balancing a chemical equation is an important skill in chemistry that helps us understand the relationships between different elements and compounds in a reaction. Here is a step-by-step guide to balancing a chemical equation.

STEP 1 Write the Chemical Equation

Start by writing the chemical equation for the reaction you want to balance. A chemical equation shows the reactants on the left side and the products on the right side, separated by an arrow. For example, let's consider the reaction between hydrogen gas and oxygen gas to form water:



STEP 2 Count the Number of Atoms on Each Side

Next, count the number of atoms of each element on both sides of the equation. This includes the atoms in both the reactants and the products. Write a list of the atoms and their counts for each element. For our example, we have:

H: 2 atoms on the left side, 2 atoms on the right side
O: 2 atoms on the left side, 1 atom on the right side

STEP 3 Choose an Element to Start Balancing

Select an element that appears in more than one molecule on either side of the equation. In our example, oxygen (O) is a good choice because it appears in two molecules on the left side and one molecule on the right side.



STEP 4 Add Coefficients

To balance the number of oxygen atoms, add a coefficient to the number in front of the molecule in the molecule that contains the element you chose. The coefficient multiplies the number of atoms in that molecule. In our example, we can balance oxygen by adding a coefficient of 2 in front of the water molecule:



Balancing Chemical Equations Packet

Sabine Zange



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