Name: _____ **Date:** _____

Balancing Chemical Equations

Balance the following chemical equations.

1.	2	Fe	+	3	H_2SO_4	$\longrightarrow$	1	$Fe_2(SO_4)_3$	+	3	H_2
2.	1	CH_4	+	2	O_2	$\longrightarrow$	1	CO_2	+	2	H_2O
3.	1	$SiCl_4(\ell)$	+	2	$H_2O(\ell)$	$\longrightarrow$	1	$SiO_2(s)$	+	4	HCl(aq)
4.	2	AgI	+	1	Na_2S	$\longrightarrow$	1	Ag_2S	+	2	NaI
5.	4	NH_3	+	5	O_2	$\longrightarrow$	4	NO	+	6	H_2O
6.	1	$FeO_3(s)$	+	3	CO(g)	$\longrightarrow$	1	$Fe(\ell)$	+	3	$CO_2(g)$
7.	1	SiO_2	+	4	HF	$\longrightarrow$	1	SiF_4	+	2	H_2O
8.	2	NaBr	+	1	Cl_2	$\longrightarrow$	2	NaCl	+	1	Br_2
9.	4	$(NH_4)_3PO_4$	+	3	$Pb(NO_3)_4$	$\longrightarrow$	1	$Pb_3(PO_4)_4$	+	12	NH_4NO_3
10.	1	$Mg(OH)_2$	+	2	HCl	$\longrightarrow$	1	$MgCl_2$	+	2	H_2O

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SJ Ball

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