Name _____ Class _____ Date _____

Chapter 6 Chemical Bonds

## WordWise

*Use the terms from the following list to fit each of the clues given below.*

| | | |
|---|---|---|
| molecule | covalent | polyatomic |
| crystals | anion | polar |
| ionic | alloy | metallic |
| formula | bond | cation |

| Clues | Vocabulary Terms |
|---|---|
| A type of bond that holds cations and anions together | _____ |
| A type of bond in which two atoms share a pair of valence electrons | _____ |
| A neutral group of atoms that are joined together by one or more covalent bonds | _____ |
| A term describing a covalent bond in which electrons are not shared equally | _____ |
| An ion that contains a covalently bonded group of atoms | _____ |
| An ion with a negative charge | _____ |
| An ion with a positive charge | _____ |
| A notation that shows what elements a compound contains and the ratio of the atoms or ions of these elements in the compound | _____ |
| Solids whose particles are arranged in a lattice structure | _____ |
| A mixture of two or more elements, at least one of which is a metal | _____ |
| A type of bond that exists between a metal cation and the shared electrons that surround it | _____ |
| The force that holds atoms or ions together | _____ |

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# Ch 6 Wordwise

Rachel Sandford

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