

Answer Key – Class 10 Science: Chemical Reactions and Equations

Section A: Multiple Choice Questions

1. c) $\text{Zn} + \text{CuSO}_4 \rightarrow \text{ZnSO}_4 + \text{Cu}$
2. b) Copper is deposited on iron
3. a) Hydrogen
4. d) All of the above
5. d) $\text{Zn} + \text{H}_2\text{SO}_4 \rightarrow \text{ZnSO}_4 + \text{H}_2$

Section B: Very Short Answer Questions

6. $\text{Ca} + 2\text{H}_2\text{O} \rightarrow \text{Ca(OH)}_2 + \text{H}_2 \uparrow$
7. A precipitation reaction is a reaction where an insoluble solid (precipitate) is formed.
Example:
 $\text{AgNO}_3 + \text{NaCl} \rightarrow \text{AgCl} \downarrow + \text{NaNO}_3$
8. Colour changes from blue to light green. Equation: $\text{Fe} + \text{CuSO}_4 \rightarrow \text{FeSO}_4 + \text{Cu}$
9. Double displacement reaction.
Balanced: $\text{BaCl}_2 + \text{Na}_2\text{SO}_4 \rightarrow \text{BaSO}_4 \downarrow + 2\text{NaCl}$
10. (i) Change in colour
(ii) Evolution of gas / Formation of precipitate

Section C: Short Answer Questions

11. A redox reaction involves both oxidation and reduction.
Example: $\text{Zn} + \text{CuSO}_4 \rightarrow \text{ZnSO}_4 + \text{Cu}$
Zn is oxidised (reducing agent), Cu^{2+} is reduced (oxidising agent).
12. On heating, lead nitrate decomposes to give lead oxide, nitrogen dioxide and oxygen.
 $2\text{Pb(NO}_3)_2 \rightarrow 2\text{PbO} + 4\text{NO}_2 + \text{O}_2$
Type: Thermal decomposition reaction.

Chapter 10 Chemical Reactions Answer Key

Judy Capra



Chapter 10 Chemical Reactions Answer Key:

Chemistry, Thermodynamics, and Reaction Kinetics for Environmental Engineers Jeff Kuo, 2024-09-25 This book aims to be the preeminent university chemistry textbook for environmental engineers. It provides undergraduate and graduate environmental engineering students with basic concepts and practical knowledge about chemistry that they would need in their professional careers. It focuses on the fundamental concepts of chemistry and its practical applications, e.g., understanding fate and transport of chemicals/pollutants in the environment, as well as the chemical/physicochemical processes applied in environmental engineering industry. This book also serves as a valuable resource for entry-level professionals to solidify their fundamental knowledge in environmental engineering chemistry. This book presents the fundamentals of chemistry with focus on the needs of environmental engineers. Explains how an understanding of chemistry allows readers a better understanding of the fate and transport of chemicals in the environment, as well as various treatment processes. Examines the fundamentals of chemical reaction equilibrium from learning the basics of thermodynamics. Presents the basic types and designs of reactors, as well as reaction kinetics.

Enzyme Kinetics: Catalysis and Control Daniel L. Purich, 2010-06-16 Far more than a comprehensive treatise on initial rate and fast reaction kinetics, this one-of-a-kind desk reference places enzyme science in the fuller context of the organic/inorganic and physical/chemical processes occurring within enzyme active sites. Drawing on 2600 references, *Enzyme Kinetics: Catalysis and Control* develops all the kinetic tools needed to define enzyme catalysis, spanning the entire spectrum from the basics of chemical kinetics and practical advice on rate measurement to the very latest work on single-molecule kinetics and mechanoenzyme force generation, while also focusing on the persuasive power of kinetic isotope effects, the design of high-potency drugs, and the behavior of regulatory enzymes. Historical analysis of kinetic principles, including advanced enzyme science, provides both theoretical and practical measurements tools. Coverage of single-molecular kinetics, examination of force-generation mechanisms, discussion of organic and inorganic enzyme reactions.

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The Elements of Qualitative Chemical Analysis Julius Stieglitz,1911 **The Retardation of Chemical Reactions** Kenneth

Claude Bailey,1937 [Chemical News and Journal of Industrial Science](#) ,1901 **Industrial and Engineering Chemistry** ,1928 **Industrial & Engineering Chemistry** ,1927 **Mathematical Modelling** Simon Serovajsky,2021-11-23

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