

**Chapter 10 test****Matching**

Match each item with the correct statement below.

- |                            |                                      |
|----------------------------|--------------------------------------|
| a. representative particle | d. percent composition               |
| b. mole                    | e. standard temperature and pressure |
| c. Avogadro's number       | f. empirical formula                 |

- \_\_\_\_\_ 1. the number of representative particles of a substance present in 1 mole of that substance
- \_\_\_\_\_ 2. an atom, an ion, or a molecule, depending upon the way a substance commonly exists
- \_\_\_\_\_ 3. the SI unit used to measure amount of substance
- \_\_\_\_\_ 4. 0°C and 1 atm
- \_\_\_\_\_ 5. the percent by mass of each element in a compound
- \_\_\_\_\_ 6. the smallest whole number ratio of the atoms in a compound

**Multiple Choice**

Identify the letter of the choice that best completes the statement or answers the question.

- \_\_\_\_\_ 7. What SI unit is used to measure the number of representative particles in a substance?
- |             |           |
|-------------|-----------|
| a. kilogram | c. kelvin |
| b. ampere   | d. mole   |
- \_\_\_\_\_ 8. How many hydrogen atoms are in 5 molecules of isopropyl alcohol, C<sub>3</sub>H<sub>7</sub>O?
- |                                     |                                      |
|-------------------------------------|--------------------------------------|
| a. $5 \times (6.02 \times 10^{23})$ | c. 35                                |
| b. 5                                | d. $35 \times (6.02 \times 10^{23})$ |
- \_\_\_\_\_ 9. All of the following are equal to Avogadro's number EXCEPT \_\_\_\_\_
- |  |
|--|
| a. the number of atoms of bromine in 1 mol Br <sub>2</sub>     |
| b. the number of atoms of gold in 1 mol Au                     |
| c. the number of molecules of nitrogen in 1 mol N <sub>2</sub> |
| d. the number of molecules of carbon monoxide in 1 mol CO      |
- \_\_\_\_\_ 10. How many moles of tungsten atoms are in  $4.8 \times 10^{23}$  atoms of tungsten?
- |                               |                               |
|-------------------------------|-------------------------------|
| a. $8.0 \times 10^{-2}$ moles | c. $1.3 \times 10^{-1}$ moles |
| b. $8.0 \times 10^1$ moles    | d. $1.3 \times 10^{-2}$ moles |
- \_\_\_\_\_ 11. How many atoms are in 0.075 mol of titanium?
- |                         |                         |
|-------------------------|-------------------------|
| a. $1.2 \times 10^{23}$ | c. $6.4 \times 10^2$    |
| b. $2.2 \times 10^{24}$ | d. $4.5 \times 10^{22}$ |
- \_\_\_\_\_ 12. How many molecules are in 2.10 mol CO<sub>2</sub>?
- |                                    |                                     |
|------------------------------------|-------------------------------------|
| a. $2.53 \times 10^{24}$ molecules | c. $3.49 \times 10^{-24}$ molecules |
| b. $3.79 \times 10^{24}$ molecules | d. $1.26 \times 10^{24}$ molecules  |

# Chapter 10 Test A

**M Walker**



**Chapter 10 Test A:**

## The Enigmatic Realm of **Chapter 10 Test A**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing in short supply of extraordinary. Within the captivating pages of **Chapter 10 Test A** a literary masterpiece penned by a renowned author, readers embark on a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting affect the hearts and minds of those that partake in its reading experience.

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