

# Atomic Structure

## Across

[3] Plays in radioactive atoms that changes a neutron to a proton or electron.

[6] A unit of mass that describes the mass of an atom or molecule.

[8] A subatomic particle that has no charge and that is found in the nucleus of an atom.

[9] An atom's central region, which is made up of protons and neutrons.

[11] The smallest unit of an element that maintains the properties of that element.

[12] A force that attracts any objects with mass.

[14] An atom that has the same number of protons/atomic number as other atoms of the same element do but that has a different number of neutrons/atomic mass.

[15] A subatomic particle that has a negative charge.

## Down

[1] The mass of an atom expressed in atomic mass units.

[2] A region around the nucleus of an atom whose electrons are likely to be found.

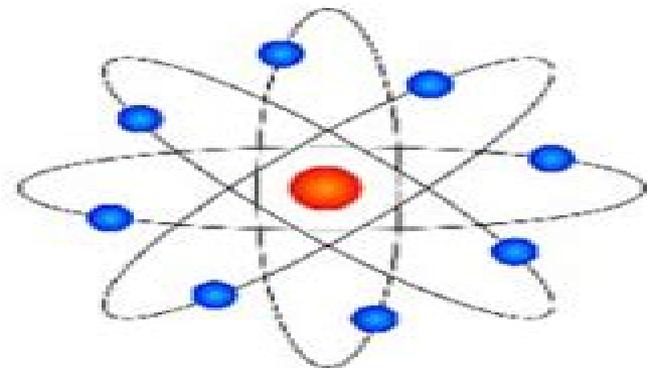
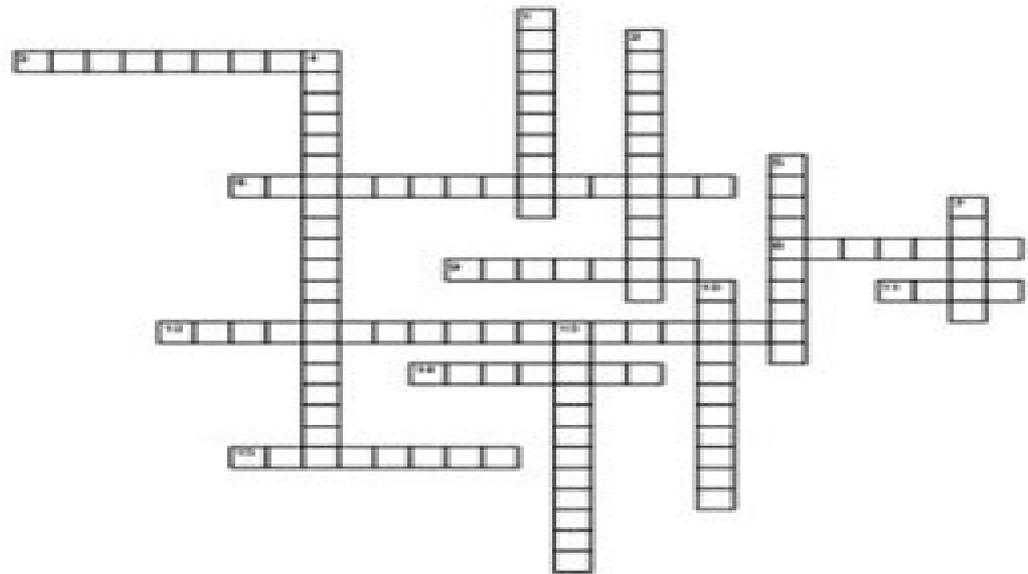
[4] A type of physical that occurs between electrically charged particles.

[5] The sum of the number of protons and neutrons in the nucleus of an atom.

[7] A subatomic particle that has a positive charge and that is found in the nucleus of an atom.

[10] The force that holds particles together in the atomic nucleus.

[13] The number of protons in the nucleus of an atom; the atom's number is the same for all atoms of an element.



# Atomic Structure Crossword Answer

**Didier Musso**



**Atomic Structure Crossword Answer:**

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