

ATOMIC THEORY TIMELINE FOLDABLE

Democritus
"Democritus Model"
*Solid round ball
-Thought nature's basic particle was an atom.
-This view was not supported by any experiments.
-Believed atoms were a round, solid ball.
400 B.C.

John Dalton
"Dalton Atomic Theory"
*Solid sphere model
-Atoms cannot be created or destroyed.
-Atoms are indivisible particles, which cannot be created or destroyed.
-Relative mass and chemical properties of atoms are constant.
-Atoms of different elements have different masses and chemical properties.

J.J. Thomson
"Plum Pudding Model"

-Negative charge distributed evenly positive charge.
-A current was passed through a cathode and a coil of something glass.
-A magnet was held to the glowing part of the cathode (called the cathode ray).
-The ray was deflected by the magnet, meaning the ray was negative.
-Thomson called this ray electrons.
1897

Ernest Rutherford
"Planetary Model"

*Nucleus of positive charge and a dense cloud of negative charge.
-Alpha particles were shot through gold foil.
-Some particles were deflected straight back.
-There had to be a small, positively charged charge in the middle of the atom.
-Rutherford called this a nucleus.
1911

Niels Bohr
"Bohr's Model"
-Electrons revolve around the nucleus in fixed orbits.
-Electrons do not radiate energy while revolving in these fixed orbits.
-The electrons' orbits are called shells.
-Electrons revolve around the nucleus in shells.
1913

Atomic Theory Timeline Project Study Guide Questions

G Psacharopoulos



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