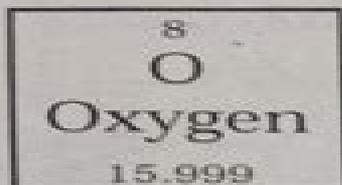


The Atoms Family
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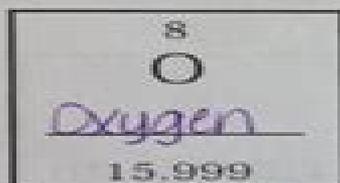
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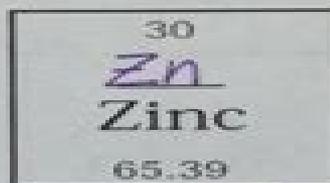
← Atomic Number
← Symbol
← Name
← Atomic Mass

Atomic number equals
the number of
protons or electrons

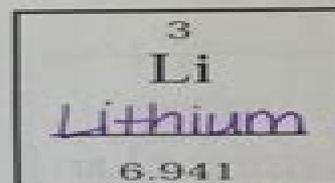
Atomic mass equals
the number of
protons + neutrons



Atomic # = 8
Atomic Mass = 16
of Protons = 8
of Neutrons = 8
of Electrons = 8



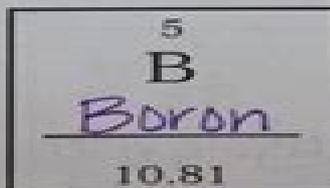
Atomic # = 30
Atomic Mass = 65
of Protons = 30
of Neutrons = 35
of Electrons = 30



Atomic # = 3
Atomic Mass = 7
of Protons = 3
of Neutrons = 4
of Electrons = 3



Atomic # = 14
Atomic Mass = 28
of Protons = 14
of Neutrons = 14
of Electrons = 14



Atomic # = 5
Atomic Mass = 11
of Protons = 5
of Neutrons = 6
of Electrons = 5



Atomic # = 35
Atomic Mass = 80
of Protons = 35
of Neutrons = 45
of Electrons = 35

Atoms Family Atomic Math Challenge Answer Key

Michal Rosen-Zvi



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