

# ATOMIC STRUCTURE

## RUTHERFORD'S MODEL

- Discovered that an atom is mostly made up of empty space
- Center of the atom is a small dense positively charged nucleus
- This attraction holds the negatively charged particles in the atom

## BOHR'S MODEL

- There are 2 main pieces of an atom:
- Nucleus
- Electrons

### THE EARLY MODEL OF THE ATOM



## WAVE-MECHANICAL MODEL

- Electrons are moving so fast that they look like a cloud. Orbitals are regions of most likely electron location



## WHAT IS TRUE TODAY...

- The nucleus is composed of 2 things:
  - PROTONS - positively charged particles
  - NEUTRONS - neutral charged particles
- Electrons are in a cloud - they are in orbitals
  - Electrons DO move around the nucleus in different patterns



Subatomic Particle	Location in the Atom	Charge	Weight
proton	center (in the nucleus)	positive	1 amu
neutron	center (in the nucleus)	neutral (no charge)	1 amu
electron	around the nucleus in clouds or orbitals	negative	Almost equal to 1/1836

### IONS

- If an atom gains or loses an electron, it is called an ion
- This gives the atom a charge:
  - Lose an electron, the atom becomes positive
  - Gain an electron, the atom becomes negative

### ISOTOPES

- An isotope is when an atom is proton or has a different number
- An isotope is still the same element, it is just a little different from every other atom of the same element
  - slightly different mass

# Atoms Study Guide

**University High School (Oakland,  
Calif.)**



## Atoms Study Guide:

Study Guide for Chemical Principles [by] Steven S. Zumdahl Steven S. Zumdahl, Paul B. Kelter, 1995 The Study Guide reflects the unique problem solving approach taken by the Chemical Principles text The new edition of the Study Guide includes many new worked out examples Atoms and the Periodic Table (Inorganic Chemistry Fast Facts) E Staff, Learn and review on the go Use Quick Review Chemistry Study Notes to help you learn or brush up on the subject quickly You can use the review notes as a reference to understand the subject better and improve your grades Easy to remember facts to help you perform better Important facts that you need to know about the Periodic Table and atomic trends Perfect study notes for all high school health sciences premed medical and nursing students **Defining the Atom (Chemistry Quick Facts)** E Staff, Learn and review on the go Use Quick Review Chemistry Study Notes to help you learn or brush up on the subject quickly You can use the review notes as a reference to understand the subject better and improve your grades Easy to remember facts to help you perform better Perfect study notes for all high school health sciences premed medical and nursing students *Atoms, Molecules, and Reactions* J. D. Laposa, 1994 *Study Guide for an Introduction to Chemistry - Atoms First* Mark Bishop, 2007 *Basic Inorganic Chemistry* F. Albert Cotton, Geoffrey Wilkinson, Paul L. Gaus, 1995-01-18 Explains the basics of inorganic chemistry with a primary emphasis on facts then uses the student's growing factual knowledge as a foundation for discussing the important principles of periodicity in structure bonding and reactivity New to this updated edition improved treatment of atomic orbitals and properties such as electronegativity novel approaches to the depiction of ionic structures nomenclature for transition metal compounds quantitative approaches to acid base chemistry Wade's rules for boranes and carboranes the chemistry of major new classes of substances including fullerenes and silenes plus a chapter on the inorganic solid state **Organic Chemistry** Marc Loudon, Jim Parise, 2015-07-01 Loudon and Parise's Organic Chemistry is known for its clear writing high standard of accuracy and creative problems This edition contains over 1 800 problems many of them new and taken directly from the scientific literature The book is used at a wide variety of schools such as UC Berkeley Caltech Colorado Cornell Duke Harvard Illinois Maryland Purdue Yale Wisconsin and many more This edition provides students with more health examples drawn from modern medical practice as well as many cutting edge topics from modern synthetic organic chemistry In addition to the printed book students can rely on Sapling Learning's online homework platform for extra learning and assessment The platform offers automatic grading an easy to use interface and instructive feedback Instructors can select from a variety of existing problem sets over 1 000 of Loudon's problems are in the platform or they can modify the questions or author them from scratch Not only does the software allow students to easily draw and interact with structures it allows them to draw entire reaction mechanisms including showing the movement of electrons with curved electron arrows *Resources in Education*, 1984 **University High School Journal**, 1927 Platers' Guide, 1922 *Fire Control II, Student Manual* California Fire Service Training Program, California. Bureau of

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