

Atomic Structure Worksheet

Label the parts of an atom on the diagram below.



4. What type of charge does a proton have?
Positive (+1)

5. What type of charge does a neutron have?
Neutral (0)

6. What type of charge does an electron have?
Negative (-1)

7. Which two subatomic particles are located in the nucleus of an atom?
Protons and neutrons

8. If an atom has 35 protons in the nucleus, how many electrons will it have orbiting the nucleus?
35

9. What is the atomic number of the atom in the diagram above?
5

10. What is the atomic mass/mass number of the atom in the diagram above?
(11+11=22) *protons and neutrons added together) 22 is the atomic mass/mass number

11. How many protons are in the nucleus of an atom with an atomic number of 15?
15 *number of protons is the same as the atomic number

12. How many electrons are in the nucleus of an atom with an atomic number of 20?
20

13. How many neutrons are in the nucleus of an atom with an atomic number of 25?
(use Periodic Table for mass)
30 Neutrons. atomic mass: 54.938 ~ 55 atomic number: 25.....55-25=30

14. What is the mass number of an atom with 3 protons, 4 neutrons, and 3 electrons?
Mass number= 6.94. (Li-Lithium has an atomic number of 3 which means it has 3 protons.

15. How many neutrons are in the nucleus of an atom that has an atomic mass of 36 and an atomic number of 25?
Neutrons= 11 (atomic mass: 36 - 25 :atomic number)

Atomic Structure Answer Key

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Atomic Structure Answer Key:

CBSE Chemistry Chapterwise Case Study Class 11 Priti Singhal, 2024-11-17 This book is structured to align with the latest syllabus and curriculum guidelines ensuring that the content is both relevant and rigorous Each chapter begins with a clear set of learning objectives providing a roadmap for students to understand what they will achieve by the end of the chapter We have included numerous diagrams illustrations and real life examples to make complex concepts more accessible and engaging

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