

### Chemistry Chapter 6 Review **KEY**

1. What is the difference between a covalent bond and an ionic bond?  
**Covalent** – share electrons to reach a full octet (Example: H—O—H)  
**Ionic** – steals electrons to reach a full octet (Example: Na<sup>+</sup> Cl<sup>-</sup>)
  
2. Determine whether the bond formed between each pair of atoms will be ionic or covalent:
  - a. K and I            **ionic**
  - b. Mg and O        **ionic**
  - c. C and Cl         **covalent**
  - d. Ca and N         **ionic**
  - e. Li and O          **ionic**
  - f. C and H           **covalent**
  
3. Then write the formula of the ionic compound. (Remember that the positive and negative charges must cancel out.)
  - a. Sodium and oxygen            Na<sup>+</sup>    O<sup>2-</sup>    **Na<sub>2</sub>O**
  - b. Magnesium and chlorine        Mg<sup>2+</sup>   Cl<sup>-</sup>    **MgCl<sub>2</sub>**
  - c. Potassium and fluorine         K<sup>+</sup>    F<sup>-</sup>     **KF**
  - d. Calcium and carbonate (CO<sub>3</sub><sup>2-</sup>)   Ca<sup>2+</sup>   CO<sub>3</sub><sup>2-</sup>   **CaCO<sub>3</sub>**  
 (Note: This is a polyatomic ion, meaning that it is a covalent compound with a charge. You will become experts on them in Unit 7.)
  
4. In your own words, define **polarity**:  
**One atom pulls more strongly on the atoms in a bond. Unequal distribution of electrons in a bond. The one Border collie pulls more strongly on the toy than the other Border collie, but instead of border collies they are atoms and instead of a chew toy it is electrons. So on and so forth.**
  
5. For each bond listed below, label the polarity by drawing an arrow in the correct direction (pointing towards the more electronegative atom).
  - a.  $\overset{\text{S}}{\text{---}} \overset{\text{O}}{\text{---}} \rightarrow$
  - b.  $\overset{\text{H}}{\text{---}} \overset{\text{Cl}}{\text{---}} \rightarrow$
  - c.  $\overset{\text{C}}{\text{---}} \overset{\text{F}}{\text{---}} \rightarrow$

# Chemistry Chapter 6 Answer Key File

**O García**



**Chemistry Chapter 6 Answer Key File:**

## Decoding **Chemistry Chapter 6 Answer Key File**: Revealing the Captivating Potential of Verbal Expression

In a period characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its capability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Chemistry Chapter 6 Answer Key File**," a mesmerizing literary creation penned by way of a celebrated wordsmith, readers set about an enlightening odyssey, unraveling the intricate significance of language and its enduring effect on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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