Chapter 2

Is Matter Around Us Pure

Multiple Choice Questions

1. Which of the following statements are true for pure substances?
   (i) Pure substances contain only one kind of particles
   (ii) Pure substances may be compounds or mixtures
   (iii) Pure substances have the same composition throughout
   (iv) Pure substances can be exemplified by all elements other than nickel
   (a) (i) and (ii)
   (b) (i) and (iii)
   (c) (iii) and (iv)
   (d) (ii) and (iii)

2. Rusting of an article made up of iron is called
   (a) corrosion and it is a physical as well as chemical change
   (b) dissolution and it is a physical change
   (c) corrosion and it is a chemical change
   (d) dissolution and it is a chemical change

3. A mixture of sulphur and carbon disulphide is
   (a) heterogeneous and shows Tyndall effect
   (b) homogeneous and shows Tyndall effect
   (c) heterogeneous and does not show Tyndall effect
   (d) homogeneous and does not show Tyndall effect

4. Tincture of iodine has antiseptic properties. This solution is made by dissolving
   (a) iodine in potassium iodide
   (b) iodine in vaseline
   (c) iodine in water
   (d) iodine in alcohol

5. Which of the following are homogeneous in nature?
   (i) ice (ii) wood (iii) soil (iv) air
   (a) (i) and (iii)
   (b) (ii) and (iv)
   (c) (i) and (iv)
   (d) (iii) and (iv)

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Camilla Rothe

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