

Test "Chemistry"

Chapter # 3

Objective part

Time: 15 min

Total marks: 12

Q.1- Every question has four possible answers, Encircle the correct option.

- i) The atomic radii of the elements in periodic table;
a) increase from left to right in a period
b) increase from top to bottom in a group
c) do not change from left to right in a period
d) decrease from top to bottom in a group
- ii) Mendeleev Periodic table was based upon the:
a) electronic configuration
b) atomic mass
c) atomic number
d) electro negativity
- iii) The amount of energy given out when an electron is added to an atom is called;
a) ionization energy
b) lattice energy
c) electron affinity
d) electro negativity
- iv) Long form of periodic table is constructed on the basis of:
a) Mendeleev postulate
b) atomic number
c) atomic mass
d) mass number
- v) Along the period which of the following decreases:
a) electro negativity
b) ionization energy
c) atomic radius
d) electron affinity
- vi) Mark the incorrect statement about the ionization energy:
a) it is measured kJmol^{-1}
b) it is the absorption of energy
c) it decreases in a period
d) it decreases in groups
- vii) Transition elements are:
a) all gases
b) all metals
c) all non metals
d) all metalloids
- viii) S-f series of inner transition elements are called:
a) lanthanides
b) actinides
c) halogens
d) alkali metals
- ix) If we move from left to right in a period the value of ionization energy:
a) remain same
b) decreases
c) increases
d) not effected
- x) Which scientist has given the idea of actives in periodic table:
a) Mendeleev
b) Al-razi
c) Newland
d) dobernier
- xi) Shielding effect is greater in atoms with greater number of:
a) proton
b) electron
c) neutron
d) positron
- xii) Which is the incomplete period in the periodic table:
a) 4th period
b) 2nd period
c) 7th period
d) 1st period

Chemistry Chapter 3 Test Questions

L Reisser

A decorative graphic element consisting of a light blue horizontal bar with a rounded right end, and a red-to-white gradient semi-circle on the right side.

Chemistry Chapter 3 Test Questions:

Chemistry Chapter 3 Test Questions Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has become much more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Chemistry Chapter 3 Test Questions**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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