Chapter 10 Study Guide- Chemical Quantities

TEST: Thursday 1/27/11

10.1 The Mole: Measurement of Matter

Define the following terms:

- mole
- Avogadro's number
- representative particle
- molar mass

1. What kind of representative particles can one calculate using Avogadro's number?

Converting Moles to # of Particles ← → Converting # of Particles to Moles

1 Mole = 6.02×10^{23} particles

therefore

$$\frac{1 \text{ mol}}{6.02 \times 10^{23} \text{ particles}} \quad \text{or} \quad \frac{6.02 \times 10^{23} \text{ particles}}{1 \text{ mol}}$$

(particles → moles)

*If you are **given particles** and want to find the **moles** you **divide** by 6.02×10^{23} particles.*

(moles → particles)

*If you are **given moles** and you want to find **particles** then you **multiply** by 6.02×10^{23} particles.*

Chemistry Chemical Quantities Study Guide

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