

Chemistry 20L
Practice Problem Set 1
Answer Key
Strong Acids and Bases

1. How many milliliters of 0.0400 M Ba(OH)₂ are required to reach the equivalence point with 15.0 mL of 0.100 HCl?

Answer: Barium hydroxide furnishes two equivalents of hydroxide in solution for each mole of barium hydroxide that you started with, and HCl donates one mole of protons/mole of acid. Thus the concentration of OH⁻ = 0.0800 N. Since equivalents of acid = equivalents of base,

$$(X\text{ mL})0.0800 = (15.0\text{ mL})(0.100\text{ N}) \text{ and } X = 18.75\text{ mL.}$$

2. What volume of 0.250 M H₂SO₄ is required to reach the equivalence point with 38.40 mL of 0.200 M NaOH?

Answer: Sulfuric acid furnishes two equivalents of protons for each mole of acid. Thus the concentration of H⁺ = 0.500 N. Since equivalents of acid = equivalents of base,

$$(X\text{ mL})0.500\text{ N} = (38.40\text{ mL})(0.200\text{ N}) \text{ and } X = 15.4\text{ mL.}$$

3. What volume of 0.120 N H₂SO₄ is required to reach the equivalence point with 24.0 mL of 0.200 N NaOH?

Answer: 40.00 mL (note both concentrations are in normality)

4. How many grams of barium hydroxide are contained in 250 mL of 0.0800 N Ba(OH)₂ solution?

Answer: Note this concentration is normality. 0.0800 N = 0.0400 M
Amount of barium hydroxide in 250 mL of 0.0400 M = 0.0100 moles or 1.71 g

4. What weight of NaOH is required to neutralize 250 mL of 0.050 M HCl?

Answer: moles of HCl = moles of NaOH = 0.0125 moles
Weight of NaOH = 0.50 g

6. A sample of an organic acid weighing 0.3446 g was titrated to a phenolphthalein end point with 41.51 mL of 0.1000 N NaOH. Calculate the equivalent weight of the acid.

Answer: equivalents of acid = equivalents of NaOH
= (0.04151 L)(0.1000 N) = 0.004151

Chemistry Practice Problem Answer Key

Julia Burdge



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Introduction to General, Organic, and Biochemistry Morris Hein, Scott Pattison, Susan Arena, Leo R. Best, 2014-01-15 The most comprehensive book available on the subject Introduction to General Organic and Biochemistry 11th Edition continues its tradition of fostering the development of problem solving skills featuring numerous examples and coverage of current applications Skillfully anticipating areas of difficulty and pacing the material accordingly this readable work provides clear and logical explanations of chemical concepts as well as the right mix of general chemistry organic chemistry and biochemistry An emphasis on real world topics lets readers clearly see how the chemistry will apply to their career

Chemistry Success in 20 Minutes a Day Michael B. McGinnis, 2005 Offers a diagnostic test and twenty lessons covering vital chemistry skills

Redefining Teacher Education and Teacher Preparation Programs in the Post-COVID-19 Era Bull, Prince Hycy, Patterson, Gerrelyn Chunn, 2021-12-17 Due to the COVID 19 pandemic teacher preparation programs modified their practices to fit the delivery modes of school districts while developing new ways to prepare candidates Governmental agencies established new guidelines to fit the drastic shift in education caused by the pandemic and P 12 school systems made accommodations to support teacher education candidates The pandemic disrupted all established systems and norms however many practices and strategies emerged in educator

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Basic Concepts of Chemistry Leo J. Malone, Theodore O. Dolter, 2012-04-13 Basic Concepts of Chemistry was originally written over thirty years ago to address the needs of general chemistry students with little or no background in chemistry Over time the text has evolved beyond purposes solely aimed at a preparatory chemistry course For some preparatory chemistry students a main sequence in general chemistry may follow but for others a semester of organic and biochemistry may follow Other students enroll to simply satisfy a basic science or chemistry requirement The text was written with a level and functionality designed to accommodate the needs of each of these varied groups of students Basic Concepts of Chemistry was designed with a flexibility that allows instructors to emphasize or omit certain clearly delineated sections The mission of the text has evolved in response to the increased diversity of students and the emphasis on outcomes assessment

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Answers to questions set at the science and art examinations Science and art department,1887 *Foundations of Chemistry* Morris Hein,Susan Arena,1996 A textbook introducing matter atomic theory ionization and other aspects of chemistry to the high school student

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