

# General Chemistry Semester 2 Study Guide

Mr. Pitzer - Fall 2018

1<sup>st</sup> Period: Monday, June 11<sup>th</sup>, 8:00am - 10:00am

2<sup>nd</sup> Period: Wednesday, June 13<sup>th</sup>, 11:00am - 1:00pm

3<sup>rd</sup> Period: Thursday, June 14<sup>th</sup>, 11:00am - 1:00pm

Use your resources wisely throughout the semester. You may not get to the end of the course by a college of other students that are less active than themselves. Maximizing your learning means the use of resources that you are provided to provide, but I expect a high level of self-motivation. Please review your notes, text, and questions for additional insight.

Make yourself a learning schedule of the semester. Consider your weaknesses as a student. Think of your school and how you can improve the learning experience. It is likely, at some stage that you will have your own study, test schedule. Use the completed schedule to identify areas that you have not yet fully mastered and require additional practice.

Be flexible, consider, and discuss a schedule. Consider, experiment, adjust, and improve it.

## Study Guide: Review List

Make sure you have the following resources necessary available and that you have access to any materials they need.

- 1. Periodic Table
- 2. Chemical Properties List
- 3. Molar Masses List
- 4. Thermodynamic Data List
- 5. Learning your ability to recognize and use correct phase changes, temperature conversions
- 6. Unit conversions (know these well) for use often
  - $10^3$  mg = 1 g =  $10^{-3}$  kg
  - 1000 mL = 1 L
  - 1000 s = 1 hr
- 7. Correctly applying stoichiometry calculations

## Final Exam: Reading Challenge

Review each book, take the following sections and answer the questions. Questions are provided below for each of the corresponding sections.

- |   |   |
|---|---|
| 1. <b>Introduction to General Chemistry</b> (101) - (101) | 2. <b>Introduction to Org. Chem</b> (101) - (101) |
| ○ 1. Introduction to the study of chemistry               | ○ 1.1. The study of chemistry                     |
| ○ 1.2. The study of chemistry                             | ○ 1.2.1. The study of chemistry                   |
| ○ 1.3. The study of chemistry                             | ○ 1.3.1. The study of chemistry                   |
| 2. <b>Introduction</b> (pp. 101 - 101) - (101)            | 3. <b>Introduction</b> (pp. 101 - 101)            |
| ○ 1.1.1.1. Introduction                                   | ○ 1.1.1.1. Introduction                           |
| ○ 1.1.1.2. Introduction                                   | ○ 1.1.1.2. Introduction                           |
| 3. <b>Thermodynamics</b> (pp. 101 - 101)                  | 4. <b>Thermodynamics</b> (pp. 101 - 101)          |
| ○ 1.1.1.1. Thermodynamics                                 | ○ 1.1.1.1. Thermodynamics                         |
| ○ 1.1.1.2. Thermodynamics                                 | ○ 1.1.1.2. Thermodynamics                         |
| 4. <b>Equilibrium</b> (pp. 101 - 101)                     | 5. <b>Equilibrium</b> (pp. 101 - 101)             |

# Chemistry Semester Study Guide For Second Semester

**Robert R. Redfield**



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