

Acids and Bases Problems

1. Rank the following solutions in order of increasing acidity (from lowest to highest):

Solution A with $\text{pH} = 6.50$

Solution C with $[\text{OH}^-] = 6.0 \times 10^{-10} \text{ M}$

Solution B with $[\text{H}_2\text{O}^+] = 3.5 \times 10^{-5} \text{ M}$

Solution D with $\text{pH} = 5.85$

_____ < _____ < _____ < _____

2. Complete the following table.

| Acid | Conjugate Base |
|-------------------------|-------------------|
| HS^- | |
| | ClO_4^- |
| H_3PO_4 | |
| | HClO_3^- |

3. Write the acid-base reactions with water for each of the following acids. Label the acid, base, conjugate acid, and conjugate base.



Acid Base Reactions Practice Problems Ap Chemistry

G Orfield



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