Self-Assessment Exercise 9.1

9n1

- **Write both forward and reverse reactions and describe macroscopic characteristics of each?**

i. $N_{2\,(g)} + 3H_{2\,(g)} \rightleftharpoons 2NH_{3(g)}$

ii. $H_{2\,(g)} + I_{2\,(g)} \rightleftharpoons 2HI_{(g)}$

Solution: i. $N_{2\,(g)} + 3H_{2\,(g)} \rightleftharpoons 2NH_{3(g)}$

Forward reaction: $N_{2\,(g)} + 3H_{2\,(g)} \longrightarrow 2NH_{3(g)}$

Reverse reaction: $2NH_{3(g)} \longrightarrow N_{2\,(g)} + 3H_{2\,(g)}$

ii. $H_{2\,(g)} + I_{2\,(g)} \rightleftharpoons 2HI_{(g)}$

Forward reaction: $H_{2\,(g)} + I_{2\,(g)} \longrightarrow 2HI_{(g)}$

Reverse reaction: $2HI_{(g)} \longrightarrow H_{2\,(g)} + I_{2\,(g)}$

Macroscopic characteristics of forward and reverse reactions:

Forward Reaction	Reverse reactions
i. It is written from left to right. **ii.** Reactants produce products. **iii.** Rate is fastest in the beginning and gradually slows down	**i.** It is written from right to left **ii.** Products produce reactants. **iii.** Its rate is zero in the beginning and gradually speeds up.

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Camilla Rothe

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