

UNIT 10: CHEMICAL REACTIONS

Chemical Reactions

Section 10.1: Reactions and Equations

For each statement, write an example of chemical reaction or process. Write an equation for each, and explain why chemical equation.

- _____ a. is a chemical reaction.
- _____ b. is a physical change (state change) process.
- _____ c. is a physical process.
- _____ d. is a chemical reaction (state change) process.
- _____ e. is a chemical reaction.
- _____ f. is a chemical reaction (state change) process.
- _____ g. is a chemical reaction (state change) process.
- _____ h. is a chemical reaction (state change) process.

Write the balanced chemical equation. Write the names of reactants, products, and state.

_____	_____	_____	_____
_____	_____	_____	_____

Write the balanced chemical equation for the reaction of hydrogen and oxygen to form water. Write the names of reactants, products, and state.

Chapter Study Guide For Content Mastery Chemical Reactions

Shasha Hu



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