

TEACHING TRANSPARENCY 32

Summary of Reaction Types

Use with Chapter 9,
Section 9.2

- For each set of reactants listed below, identify the type of reaction that the reactants might undergo. List as many reaction types as may apply. Assume that all the reactants for the reaction are listed.
 - a compound and an element _____
 - two compounds _____
 - one compound _____
- For each set of reaction products listed below, identify the type of reaction that might have formed the products. List as many reaction types as apply. Assume that all the products for the reaction are listed.
 - a compound and an element _____
 - two compounds _____
 - one compound _____
- Classify each of the following examples according to the type of reaction involved. List as many reaction types as may apply.
 - A match burns.

 - The carbonic acid found in soft drinks breaks down into bubbles of carbon dioxide and water.

 - Phosphorous and oxygen react rapidly, forming diphosphorous pentoxide.

 - An iron nail is placed into a copper sulfate solution. Copper metal appears on the nail.

 - The acid in baking powder reacts with baking soda (NaHCO_3), forming carbon dioxide gas and other products.

 - Water and sulfur trioxide react to form sulfuric acid.

 - Copper wire is placed in a silver nitrate solution. The solution turns blue, which is the color of the copper ion, and solid silver forms on the wire.

Chapter Transparency 21

Albert A Gayle



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