

General Chemistry Study Guide

Chapters 4-6

Chapter 4

Key Concepts:

-Atomic Structure: The chapter will cover the basics of atomic structure, including the nature of the atom, the types of subatomic particles, and the arrangement of these particles in an atom.

-Electron Configuration: The chapter will explain how electrons are arranged in an atom and how this affects the atom's properties.

-The Periodic Table: The chapter will introduce the periodic table and explain how it is used to predict the properties of elements.

Important Events:

-The Discovery of the Atom: The chapter will cover the history of the atom, including its discovery by Democritus and its further development by scientists such as John Dalton and J.J. Thomson.

-The Development of the Periodic Table: The chapter will trace the history of the periodic table, from its beginnings with the work of Dmitri Mendeleev to its modern form.

Notable Figures:

-Democritus: Democritus was a Greek philosopher who is credited with being the first to propose the existence of atoms.

-John Dalton: Dalton was an English chemist who developed the atomic theory of matter.

-J.J. Thomson: Thomson was a British physicist who discovered the electron.

-Dmitri Mendeleev: Mendeleev was a Russian chemist who developed the periodic table of elements.

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RC Schank

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