

Physical and Chemical Changes and Properties of Matter Worksheet

Classify the following as chemical change (cc), chemical property (cp), physical change (pc), or physical property (pp).

- | | |
|-------------------------------------|-------------------------------|
| 1. <u>PP</u> Heat conductivity | 8. <u>CP</u> Combustible |
| 2. <u>CC</u> Silver tarnishing | 9. <u>PC</u> Water freezing |
| 3. <u>PC</u> sublimation | 10. <u>CC</u> Wood burning |
| 4. <u>PC</u> magnetizing steel | 11. <u>CP</u> Acid resistance |
| 5. <u>PP</u> length of metal object | 12. <u>PP</u> Brittleness |
| 6. <u>PC</u> shortening melting | 13. <u>CC</u> Milk souring |
| 7. <u>CC</u> exploding dynamite | 14. <u>CC</u> baking bread |

Identify the following as being true or false to the left of the sentence.

- T 15. A change in size or shape is a physical change.
T 16. A chemical change means a new substance with new properties was formed.
F 17. An example of a chemical change is when water freezes.
T 18. When platinum is heated, then cooled to its original state, we say this is a physical change.
F 19. When milk turns sour, this is a physical change because a change in odor does not indicate a chemical change.
T 20. When citric acid and baking soda mix, carbon dioxide is produced and the temperature decreases. This must be a chemical change.

Identify each of the following as a physical or chemical change.

21. CC You leave your bicycle out in the rain and it rusts.
22. PC A sugar cube dissolves.
23. CC Scientist break-up water into oxygen and hydrogen gas.
24. CC Burning coal for a barbecue.
25. PC Trimming a bush because it has grown too tall.

Classifying Matter Worksheet

Classify each of the following substances as an element, a compound, a solution (homogeneous mixture), or a heterogeneous mixture.

- | | | |
|---|--|--|
| 1. Sand
heterogeneous mixture | 2. Salt
Compound | 3. Pure Water
compound |
| 4. Soil
heterogeneous mixture | 5. Soda just opened
homogenous mixture | 6. Pure air
homogenous mixture |

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William Henry Howell



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