

Assessment

Chapter 12 Practice Test**Chapter: Solutions**

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- _____ 1. Agitation prevents settling in a(n)
a. alloy.
b. homogeneous mixture.
c. suspension.
d. gaseous mixture.
- _____ 2. All of the following are heterogeneous mixtures *except*
a. whole wheat bread.
b. granite.
c. tap water.
d. an oil-water mixture.
- _____ 3. What is the concentration of a 100. mL aqueous solution that contains 1.00 g KCl (molar mass = 74.55 g/mol)?
a. 1.34 M KCl
b. 0.134 M KCl
c. 0.0134 M KCl
d. 0.001 34 M KCl
- _____ 4. To determine the molarity of an HCl solution, you need to know the number of
a. grams of HCl in 1×10^6 g of solution.
b. moles of HCl dissolved in the total moles of solution.
c. moles of HCl in 1 L of solution.
d. moles of HCl dissolved in 1 kg of solvent.
- _____ 5. What type of solute-solvent combination is carbon dioxide in water?
a. gas-liquid
b. liquid-gas
c. liquid-liquid
d. cannot be determined
- _____ 6. What is the molarity of a solution that contains 0.202 mol KCl (molar mass = 74.55 g/mol) in 7.98 L of solution?
a. 0.0132 M KCl
b. 0.0253 M KCl
c. 0.459 M KCl
d. 1.36 M KCl

Chapter 12 Practice Test

Mark Steyvers



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