

## Stoichiometry Worksheet # 1

1) Glucose is used as a source of energy by the human body. The overall reaction in the body is



Calculate the number of moles of oxygen needed to oxidize 12.5 g of  $\text{C}_6\text{H}_{12}\text{O}_6$ .

2) Ammonia is synthesized from hydrogen and nitrogen according to the following equation.



If an excess of nitrogen is reacted with 3.41 moles of hydrogen gas, how many grams of ammonia ( $\text{NH}_3$ ) can be produced?

3) Assume that in the decomposition of potassium chlorate,  $\text{KClO}_3$ ,



80.5 moles of  $\text{O}_2$  form. How many grams of potassium chlorite, the other product would be formed?

4) In a single displacement reaction,



9.23 moles of aluminum react with excess hydrochloric acid. How many moles of hydrogen will be produced?

5) The compound "Cisplatin"  $\text{PtCl}_2(\text{NH}_3)_2$ , has been found to be effective in treating some types of cancer. It can be synthesized using the following reaction



A. How many moles of "cisplatin" can be produced from 2.50 g  $\text{K}_2\text{PtCl}_4$ ?

B. How many moles of  $\text{NH}_3$  would be needed if 3.8 moles of  $\text{K}_2\text{PtCl}_4$  were present?

# Chapter 12 Stoichiometry Section Assessment Answers

**Yijin Wang**



**Chapter 12 Stoichiometry Section Assessment Answers:**

## Unveiling the Energy of Verbal Artistry: An Mental Sojourn through **Chapter 12 Stoichiometry Section Assessment Answers**

In some sort of inundated with displays and the cacophony of immediate communication, the profound energy and mental resonance of verbal art frequently disappear in to obscurity, eclipsed by the regular barrage of noise and distractions. However, nestled within the musical pages of **Chapter 12 Stoichiometry Section Assessment Answers**, a charming function of literary beauty that pulses with natural feelings, lies an unforgettable journey waiting to be embarked upon. Published by a virtuoso wordsmith, this magical opus instructions readers on a psychological odyssey, lightly exposing the latent possible and profound influence stuck within the complicated web of language. Within the heart-wrenching expanse of the evocative evaluation, we shall embark upon an introspective exploration of the book is central themes, dissect its interesting publishing type, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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