

CHAPTER 19 REVIEW

Oxidation-Reduction Reactions

SECTION 1

SHORT ANSWER Answer the following questions in the space provided.

1. _____ All the following equations involve redox reactions *except*



2. Assign the correct oxidation number to the individual atom or ion below.

_____ a. Mn in MnO_2

_____ b. S in S_8

_____ c. Cl in CaCl_2

_____ d. I in IO_3^-

_____ e. C in HCO_3^-

_____ f. Fe in $\text{Fe}_2(\text{SO}_4)_3$

_____ g. S in $\text{Fe}_2(\text{SO}_4)_3$

3. In each of the following half-reactions, determine the value of x .



_____ d. Which of the above half-reactions represent reduction processes?

4. Give examples, other than those listed in Table 1 on page 631 of the text, for the following:

_____ a. a compound containing H in a -1 oxidation state

_____ b. a peroxide

_____ c. a polyatomic ion in which the oxidation number for S is $+4$

_____ d. a substance in which the oxidation number for F is not -1

Chapter 19 Practice Problems Oxidation Reduction Reactions Answers

Guillaume Favre



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