

Colligative Properties, Homework 13

Table 1: Constants

Name	Formula	$k_b \frac{^\circ\text{C}}{m_{\text{particles}}}$	$k_f \frac{^\circ\text{C}}{m_{\text{particles}}}$
Water	H ₂ O	0.512	1.86
Naphthalene	C ₁₀ H ₈	*****	6.9
Ethyl alcohol	C ₂ H ₆ O	1.22	1.99
Acetic acid	HC ₂ H ₃ O ₂	3.07	3.90
Benzene	C ₆ H ₆	2.53	4.90
Nitrobenzene	C ₆ H ₅ NO ₂	5.24	7.00

- How many particles are generated when each of the following compounds are dissolved in water. a) FeCl₂, b) NaCl, c) C₆H₁₂O₆ (glucose), d) Cr₂(SO₄)₃, e) CH₃NH₂ (methyl amine)
- Calculate the a) freezing point and b) boiling point when 78.75 g NaCl are dissolved in 275 g of water.
- Antifreeze in cars is usually made by mixing ethylene glycol, HOCH₂CH₂OH, with water to make a 50.0% ~~volume/volume~~ solution. If the specific gravity of ethylene glycol is 1.113, calculate a) the boiling and b) the freezing point of this solution. Assume that water is the solvent.
- Using the solution made in problem 3, calculate the a) k_b and b) k_f for ethylene glycol assuming it is the solvent.
- Calculate the molecular mass of ethylene glycol if a 150 g sample is placed into 250 g of ethyl alcohol, which has a boiling point of 78°C and the resulting mixture has a boiling point of 89.8°C.

Colligative Properties Homework Answer

Daniela Niemeyer



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